

SL-37. Soft Electrophiles-5. Nitriles

Covalent inhibitors often possess an electrophilic "warhead" which interacts with a nucleophilic residue at the binding site under certain physiological conditions; examples of such electrophilic groups include acrylamides, epoxides, nitriles, and electron-deficient ketones.

Nitrile-containing molecules in which a nitrile group is attached to an aromatic core are quite widespread among pharmaceuticals and pesticides [2]. Due to the presence of the highly-polarized triple bond, nitriles can interact with Ser or Cys

residues of the proteins forming covalent adducts. Several nitrilecontaining molecules have been developed as reversible covalent inhibitors of serine and cysteine proteases.

At ASINEX, we have created a library of heteroaromatic nitrile derivatives which have kinase-oriented pharmacophoric elements. These compounds are useful for the discovery of novel kinase inhibitors with improved safety, target selectivity, and pharmacokinetic properties.

Signature Library 37

Formats	Supplementary Information
80 compounds per plate	SL#37_Soft_Electrophiles_CN.sdf
0.1 mg; 1 mg; 2 mg dry film/powder	
0.1 µmol; 1 µmol DMSO solutions	

References:

- 1. Drug Discov Today. 2015 Sep;20(9):1061-73. doi: 10.1016/j.drudis.2015.05.005
- 2. *J Med Chem*. 2010 November 25; 53(22): 7902–7917. doi:10.1021/jm100762r.

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