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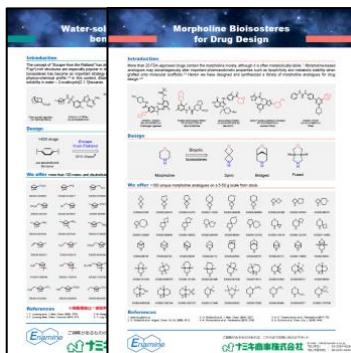


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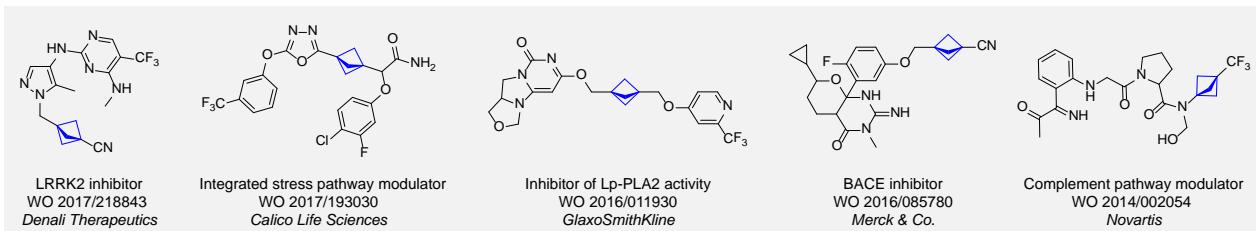
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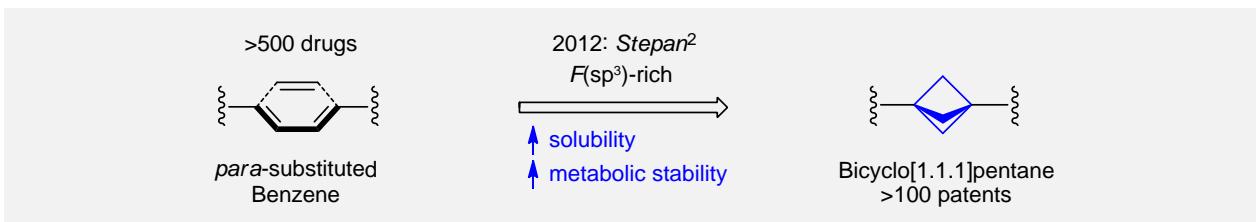
Saturated Bioisosteres of *para*-substituted Benzenes

Introduction

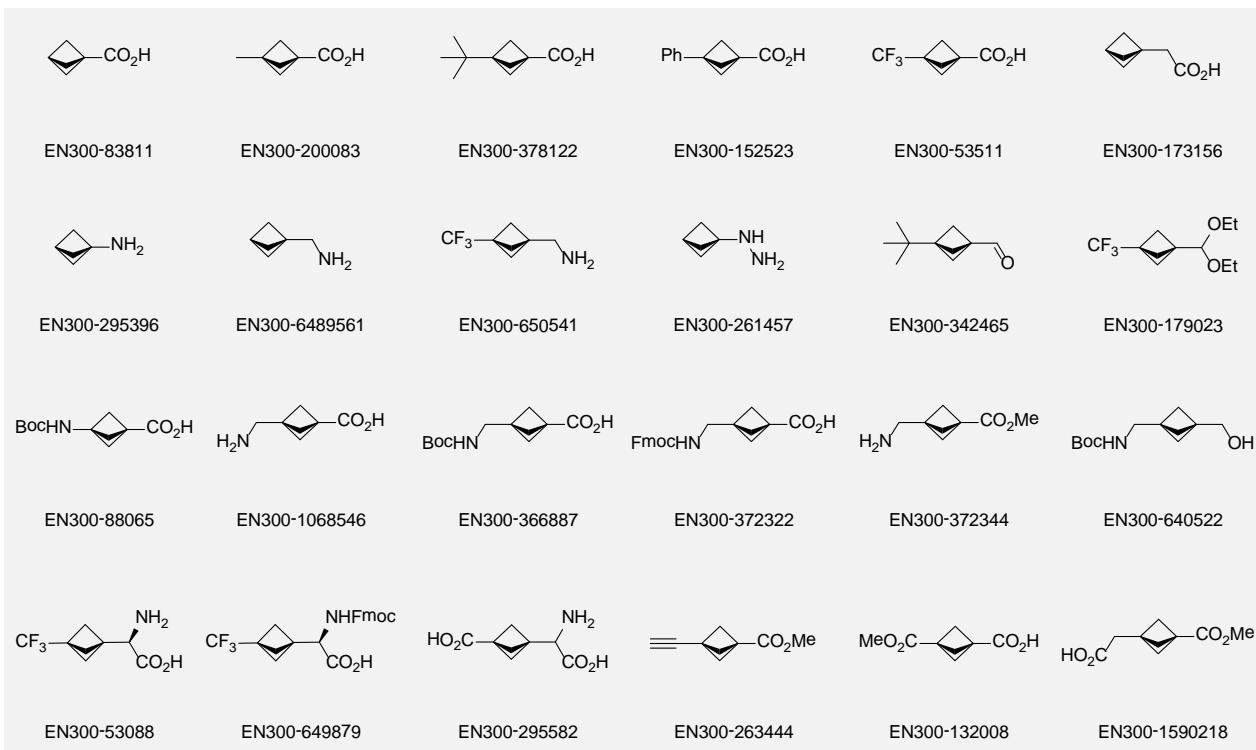
The residue of benzene comprises to the structure of more than 500 FDA-approved drugs.¹ In 2012, Stepan and coworkers showed that bicyclo[1.1.1]pentane skeleton could act as a saturated “*nonclassical phenyl ring bioisostere*” in the design of a γ -secretase inhibitor.² Since then, the core of bicyclo[1.1.1]pentane is often used in the design of analogues of natural compounds,³ peptide studies,^{4,5} medicinal chemistry,^{6,7} and supramolecular chemistry.⁸ Herein we have designed and synthesized a library of saturated mimics of the *para*-benzene ring for drug design.



Design



We offer



References

1. R. D. Taylor et al. *J. Med. Chem.* **2014**, *57*, 5845.
2. A. F. Stepan et al. *J. Med. Chem.* **2012**, *55*, 3414.
3. Y. L. Goh et al. *J. Am. Chem. Soc.* **2016**, *138*, 1698.
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7. N. D. Meason et al. *ACS Med. Chem. Lett.* **2017**, *8*, 43.
8. A. M. Dilmac et al. *ANIE* **2017**, *56*, 5684.



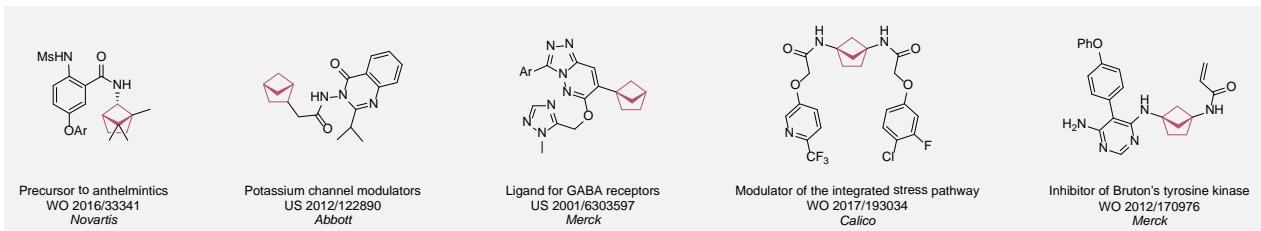
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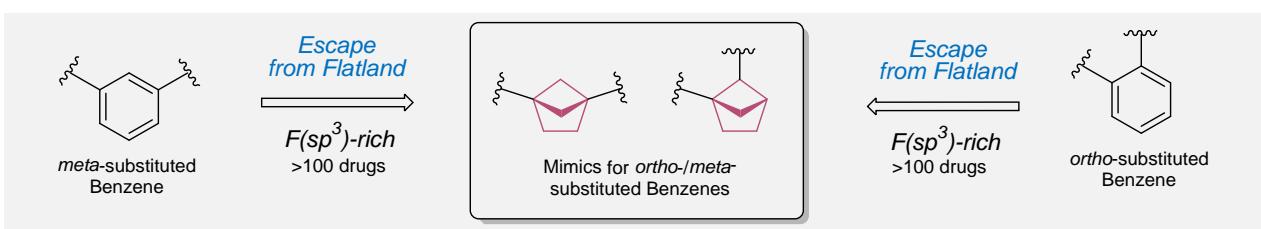
Saturated Bioisosteres of *ortho*-/*meta*-substituted Benzenes

Introduction

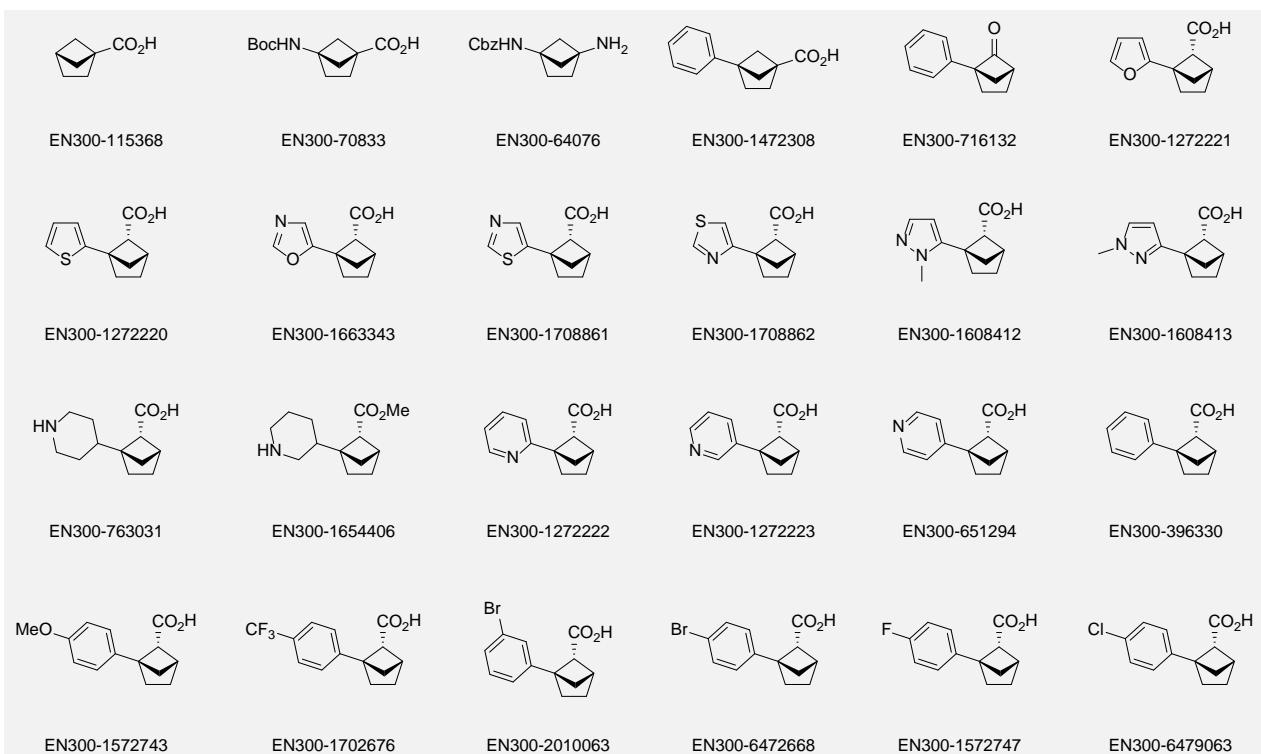
The fragment of benzene comprises to the structure of more than 500 FDA-approved drugs.¹ In 2012, Stepan and coworkers showed that bicyclo[1.1.1]pentane skeleton could act as a saturated “nonclassical phenyl ring bioisostere”.²⁻⁶ Adding one carbon atom gives the closest homologue – bicyclo[2.1.1]hexane. The lack of the practical synthetic approaches restricts the common use of bicyclo[2.1.1]hexanes in chemistry. Herein we have designed and synthesized a library of saturated mimics of the *ortho*- and *meta*-benzene ring for drug design.



Design



We offer



References

- R. D. Taylor et al. *J. Med. Chem.* **2014**, *57*, 5845.
- A. F. Stepan et al. *J. Med. Chem.* **2012**, *55*, 3414.
- Y. L. Goh et al. *J. Am. Chem. Soc.* **2016**, *138*, 1698.
- P. K. Mykhailiuk et al. *ANIE* **2006**, *45*, 5659.
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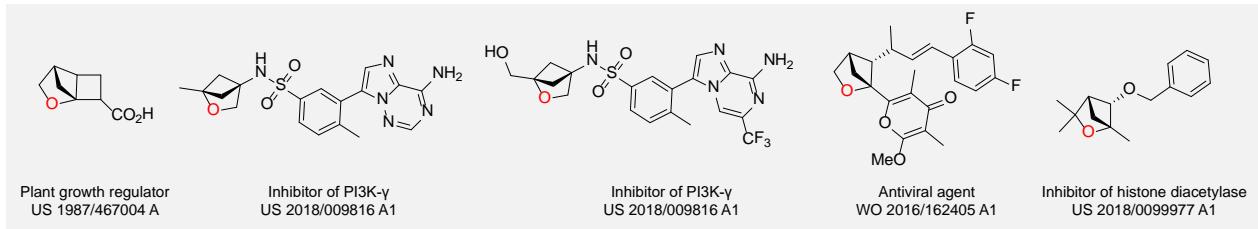
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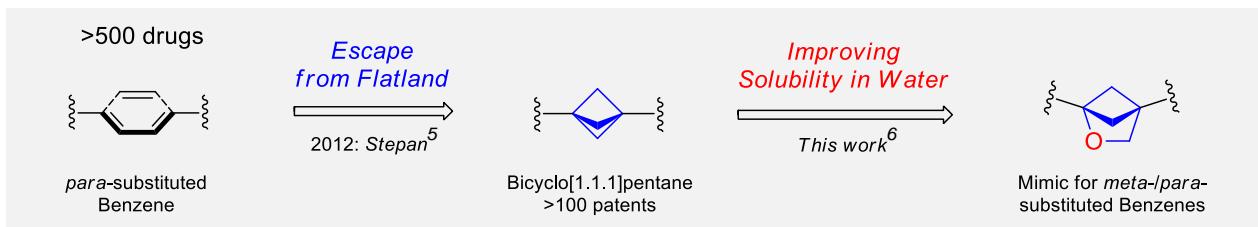
Water-soluble non-classical benzene mimics

Introduction

The concept of “*Escape from the Flatland*” has already gained considerable attention in medicinal chemistry. Nowadays small F(sp³)-rich structures are especially popular in drug discovery projects. In particular, replacing benzene rings with saturated bioisosteres has become an important strategy to obtain novel patent-free molecules with improved biological activity and physico-chemical profile.¹⁻⁶ In this context, *Enamine* offers a new generation of saturated benzene mimics with improved solubility in water – 2-oxabicyclo[2.1.1]hexanes.



Design



We offer more than 100 mono- and disubstituted benzene mimics with improved water solubility from stock on a 5-10 g scale:



References

1. F. Lovering et al. *J. Med. Chem.* **2009**, 6752.
2. F. Lovering *Med. Chem. Commun.* **2013**, 515.

3. M. Westphal et al. *ChemMedChem.* **2015**, 461.
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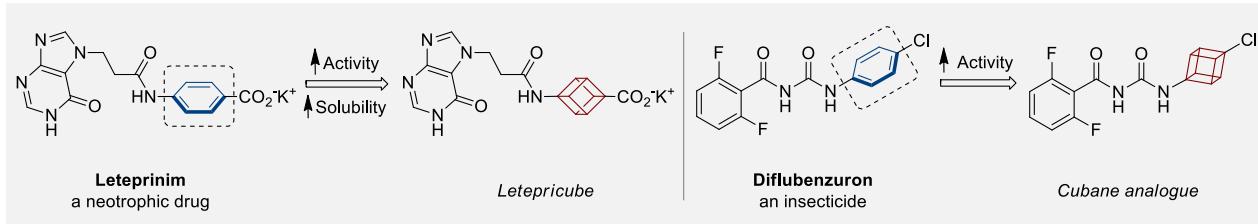
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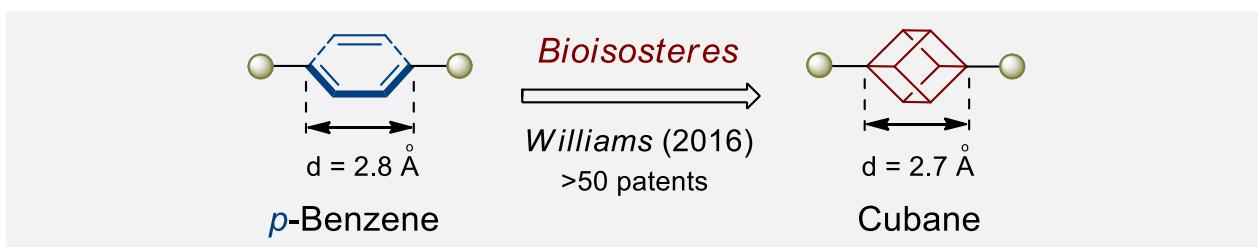
Cubanes for Medicinal Chemistry

Introduction

In 1992, Eaton predicted a high potential of cubane in a pharmaceutical research as a bioisostere of benzene, based on their similarity in size. In 2016, Williams and collaborators showed that replacing a benzene ring in the neurotropic compound *Leteprinim* with cubane beneficially affected activity and solubility of the parent compound. The cubane analogue significantly outperformed pesticide *Diflubenzuron*. Since then the cubane-containing building blocks have been playing an important role in medchem projects, as mimics for the *para*-substituted benzene ring. In this context, Enamine offers a library of cubane-containing building blocks for drug design.¹⁻⁶



Concept



We offer: cubane-containing building blocks from stock on a 5-10 g scale.

EN300-7365525	EN300-93169	EN300-7492225	EN300-6750908	EN300-6497484	EN300-7457217	EN300-1705874
EN300-6729668	EN300-7466659	EN300-26620086	EN300-26620085	EN300-22712991	EN300-26976616	EN300-7442138
EN300-22913960	EN300-7435152	EN300-88072	EN300-649920	EN300-156445	EN300-136711	EN300-93166
EN300-7457033	EN300-7459978	EN300-1698363	EN300-115528	EN300-6737755	EN300-1585736	EN300-1585726
EN300-7425276	EN300-7426785	EN300-1081961	EN300-7431073	EN300-7432402	EN300-1081965	EN300-1081966

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1. B. A. Chalmers et al. *Angew. Chem. Int. Ed.* **2016**, 3580.
2. J. Wlochal et al. *Synlett*, **2016**, 919.
3. J. Wlochal et al. *Org. Lett.* **2014**, 4094.
4. M. J. Falkiner et al. *Org. Process. Res. Dev.* **2013**, 1503.
5. P. K. Mykhailiuk. *Org. Biomol. Chem.* **2019**, 17, 2839.
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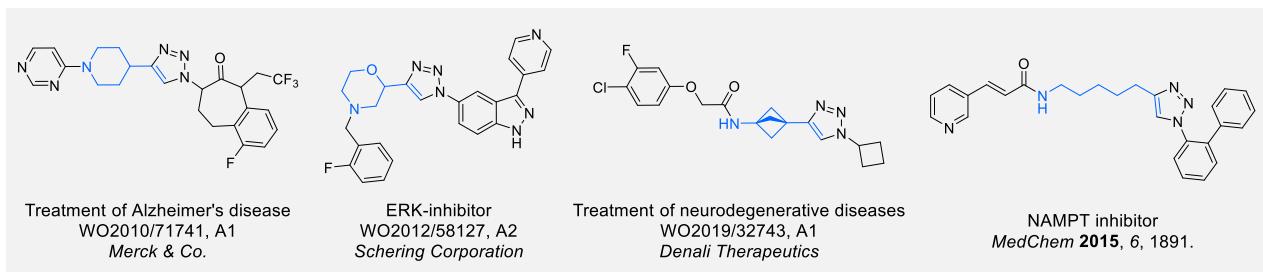
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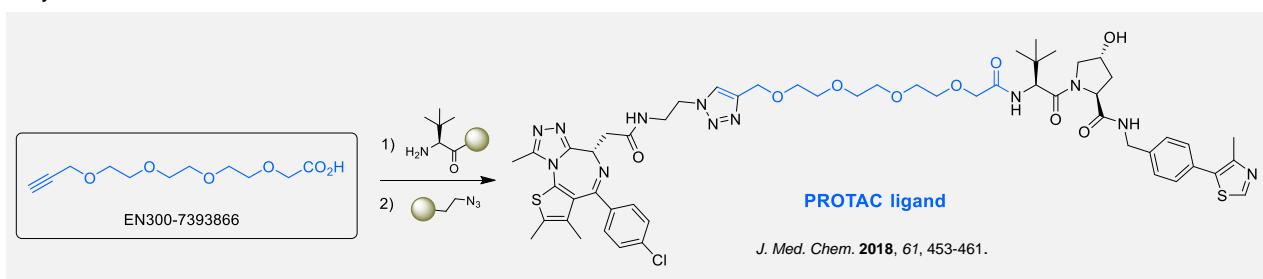
Alkyne-containing Linkers

Introduction

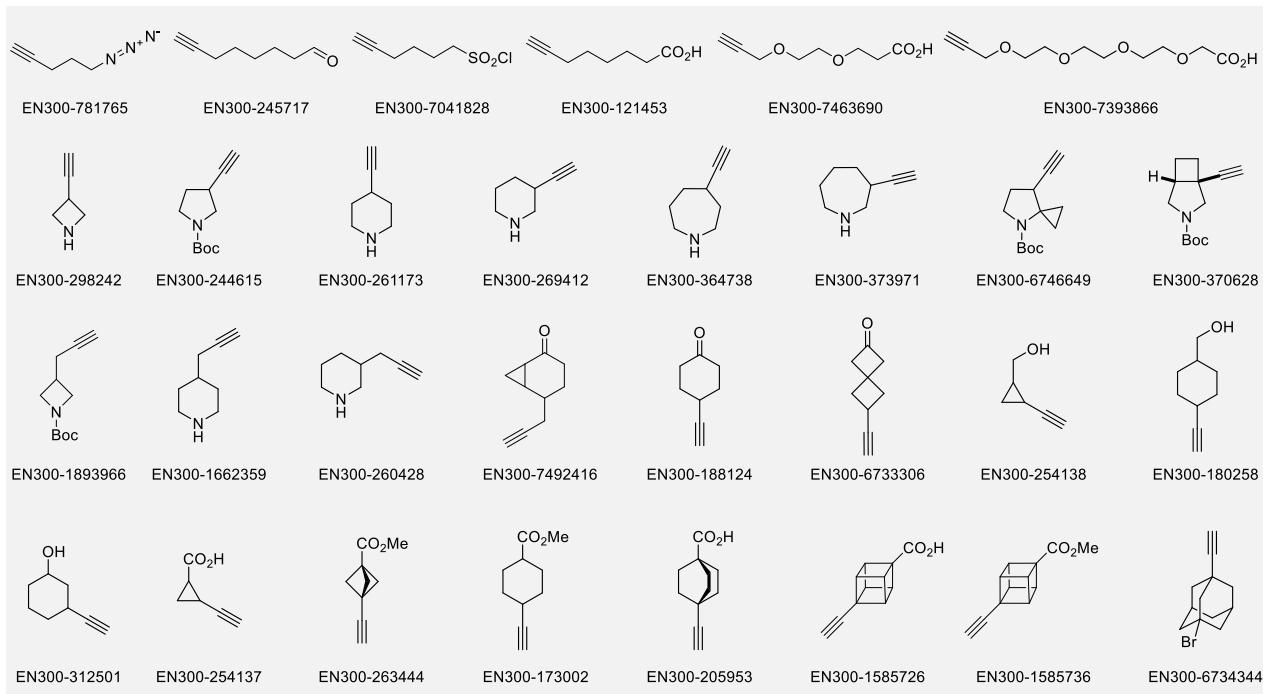
Acetylenes are used in the Huisgen azide-alkyne 1,3-dipolar cycloaddition reaction – “click chemistry”. 1,2,3-Triazole function formed by click reaction between an azide and alkyne bears a physicochemical resemblance to the amide bond. Using “click chemistry” one can efficiently connect complex molecules such as ligands and tool compounds.¹⁻⁶



In this context, Enamine offers a library of unique functionalized alkynes that can be used as linker compounds, for example, in synthesis of PROTACs.



We offer: >100 unique acetylenes on a 5-50 g scale from stock.



References

1. H. C. Kold et al. *Angew. Chem. Int. Ed.* **2001**, 2004.
2. L. H. Jones et al. *Angew. Chem. Int. Ed.* **2012**, 6320.
3. P. Thirumurugan et al. *Chem. Rev.* **2013**, 4905.
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5. N. M. Meghani et al. *Drug Discov. Today.* **2017**, 22, 1604.
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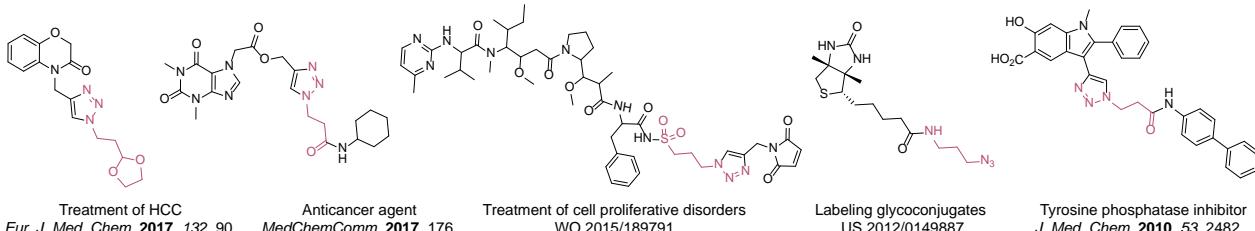
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Azide-linkers for Drug Design

Introduction

Organic azides became enormously popular for their participation in the Cu(I)-catalyzed Huisgen azide-alkyne 1,3-dipolar cycloaddition reaction – “click chemistry”. 1,2,3-triazole function formed by click reaction between an azide and alkyne bears a physicochemical resemblance to the amide bond. Besides, “click chemistry” involves functionalities that can be introduced in small molecules and into specific locations in biomolecules. “Click chemistry” continues to gain popularity and is used in a variety of research fields with significant contributions to the fields of bioconjugation and drug discovery.



Advantages

Aliphatic functionalized azides

- ◀ Wide in scope.
- ◀ Form stable products.
- ◀ Give very high yields.

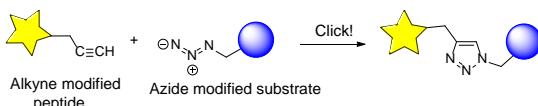
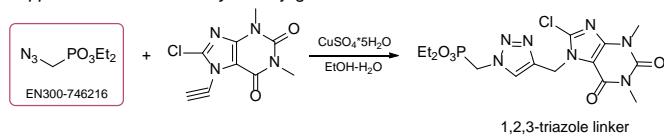


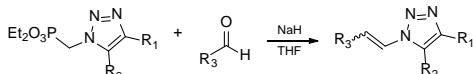
Figure 1. Click reaction forming a triazole link in peptide.

- ◀ The presence of the azide-group and a functional group allows the molecule to be modified before or after the click reaction.

Copper-mediated azide-alkyne conjugation



The Homer-Wadsworth-Emmons olefination of the triazoles



We offer

Over 100 different building blocks in multi gram amounts in stock. We also have designed a library azide-containing building blocks for drug discovery programs. These molecules can be synthesized upon request within 4-6 weeks.

	N ₃ -CO ₂ H EN300-69260	N ₃ -C(=O)Me EN300-254265	N ₃ -CO ₂ iBu EN300-72572	N ₃ -COCl EN300-136927	N ₃ -PO ₃ Et ₂ EN300-746216	N ₃ -CN EN300-122223
	N ₃ -CH ₂ CO ₂ H EN300-108329	N ₃ -CH ₂ SO ₂ Cl EN300-125654	N ₃ -CH ₂ SO ₂ F EN300-727933	N ₃ -Cyclo-COOH EN300-1725172	N ₃ -Cyclo-COOH EN300-1726255	N ₃ -CH ₂ Br EN300-93799
	N ₃ -CH ₂ CH ₂ NH ₂ EN300-54082	N ₃ -CH ₂ CH ₂ NHBoc EN300-108020	N ₃ -CH ₂ OH EN300-54800	N ₃ -Cyclo-NH EN300-259527	N ₃ -Cyclo-NBoc EN300-81032	N ₃ -Cyclo-NH EN300-257805
	N ₃ -Cyclo-NH ₂ EN300-1725170	N ₃ -Cyclo-NHBoc EN300-66132	N ₃ -Cyclo-NH ₂ EN300-1725171	N ₃ -Cyclo-NHBoc EN300-1726264	N ₃ -Cyclo-NH ₂ EN300-1725178	N ₃ -Cyclo-NBoc EN300-1664352
	N ₃ -CH ₂ CH ₂ CO ₂ H EN300-185977	N ₃ -CH ₂ CH ₂ CHO EN300-1699836	N ₃ -CH ₂ CH ₂ CO ₂ Me EN300-185976	N ₃ -CH ₂ CH ₂ SO ₂ Cl EN300-139628	N ₃ -CH ₂ CH ₂ SO ₂ F EN300-224056	N ₃ -CH ₂ CH ₂ BPin EN300-206659
	N ₃ -CH ₂ CH ₂ NH ₂ EN300-54083	N ₃ -CH(CH ₃)NH ₂ EN300-255958	N ₃ -CH ₂ OH EN300-54802	N ₃ -Cyclo-NH EN300-261853	N ₃ -Cyclo-COOH EN300-1720529	N ₃ -Cyclo-NH ₂ EN300-1726266
						N ₃ -Cyclo-NBoc EN300-265675



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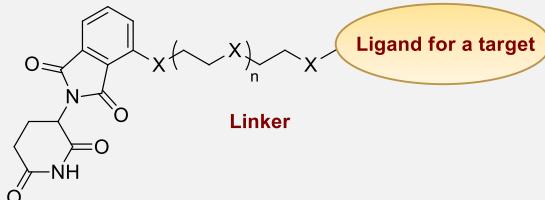
Building blocks and linkers for PROTAC synthesis

Introduction

Proteolysis targeting chimeras (PROTACs) is the recently emerged field in drug discovery. This new approach works through the activation of the ubiquitin-proteasome system to remove disease-causing proteins. This new modality of therapeutic intervention requires new chemistry and new approaches to synthesize functional PROTAC molecules. Several recent papers of Oprea and Cravatt examined chemical space and ligandable proteome to evaluate a state of the targeted human genome. Herein we offer known and novel building blocks (E3 ligase ligands, ligands with linkers, linkers) for the synthesis of PROTACs.

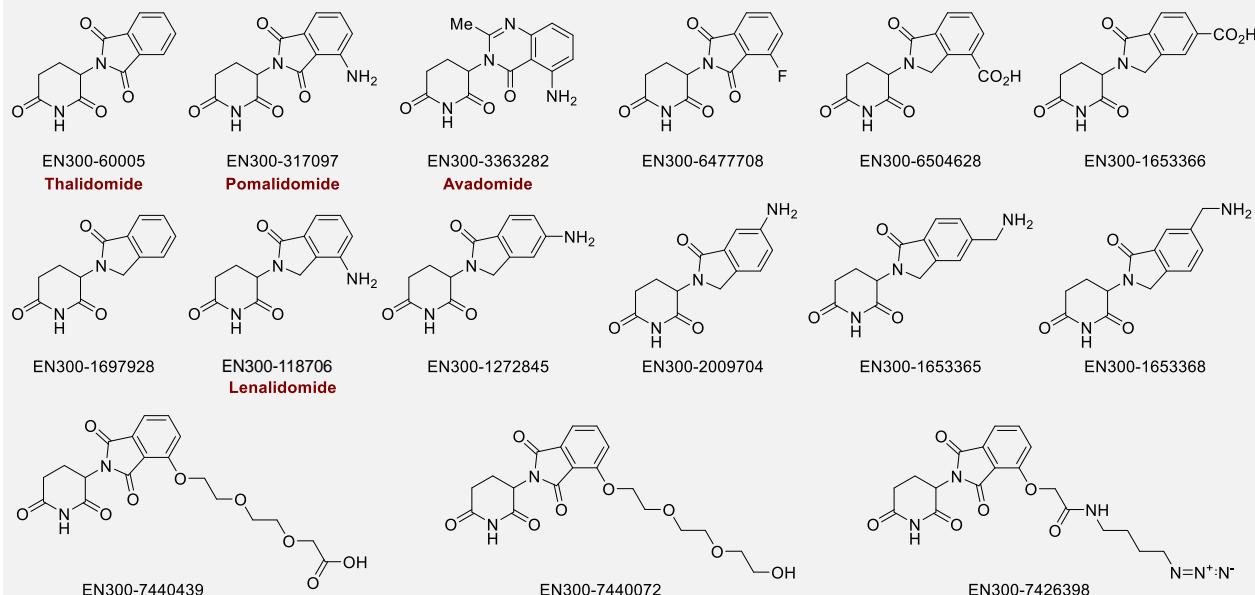
PROTAC model:

E3 ligase ligands
(Thalidomide)



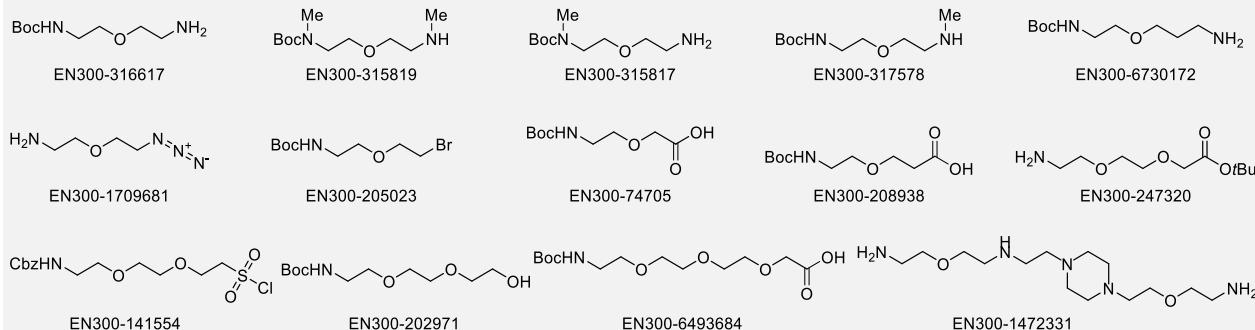
Ligands for E3 ligases (Cereblon, VHL, MDM2 etc) from stock:

over 100 ligands:



PEG-linkers from stock:

over 50 linkers:



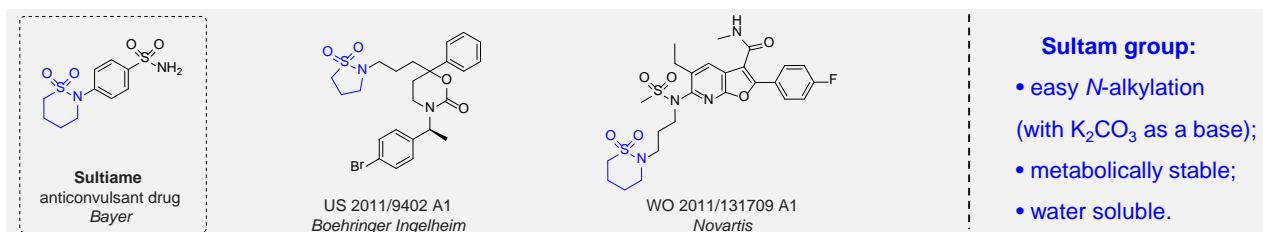
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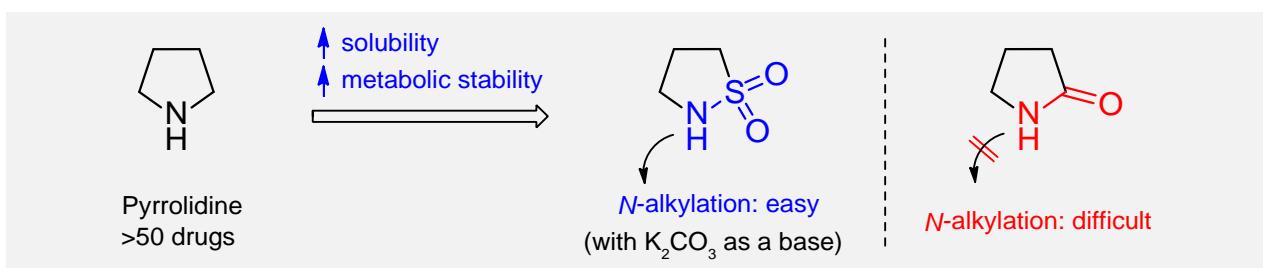
Cyclic Sulfonamides for Drug Design

Introduction

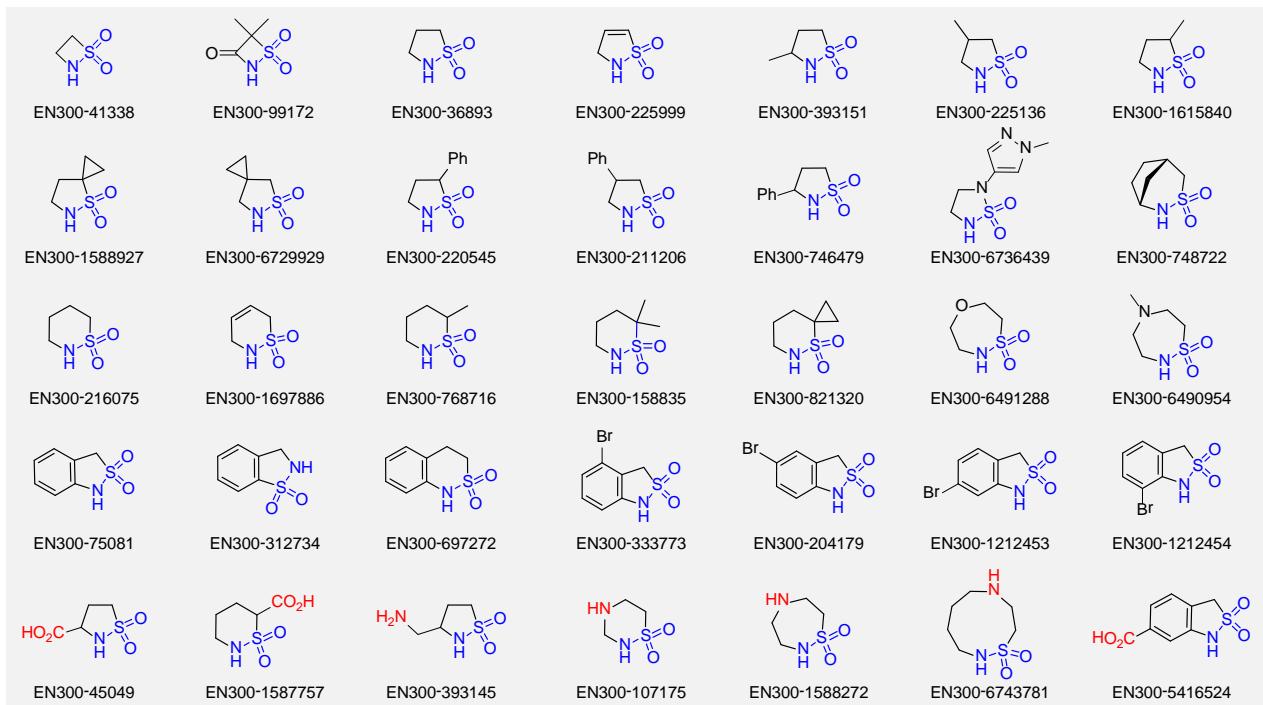
Sulfonamides are popular in drug discovery: more than 100 FDA-approved drugs on the market are sulfonamide-containing. Bioactive cyclic sulfonamides (sultams) include the anticonvulsant *Sultame* (Bayer) and the anti-inflammatory drug *Piroxicam* (Pfizer).¹⁻⁵ Mostly, the *N*-aryl substituted sultams are synthesized from the corresponding anilines. Herein, we present a library of aliphatic sultams that can be easily alkylated at the *N*-atom. These compounds can be considered as water-soluble mimics of common cyclic amines – pyrrolidines, piperidines, etc.



Design



We offer >50 unique sultams on 5-50 g scale in stock.



References

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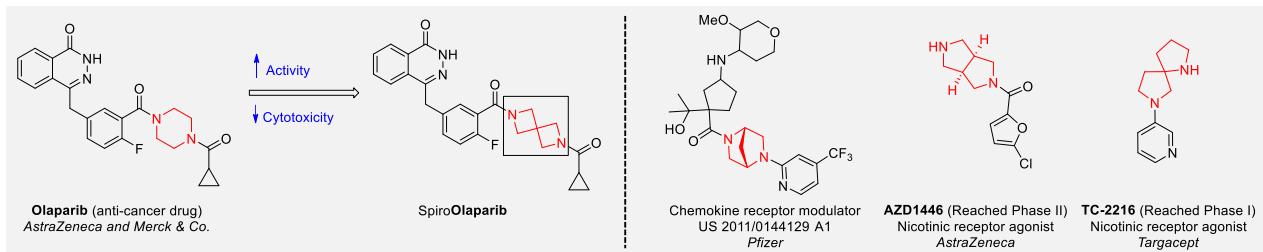
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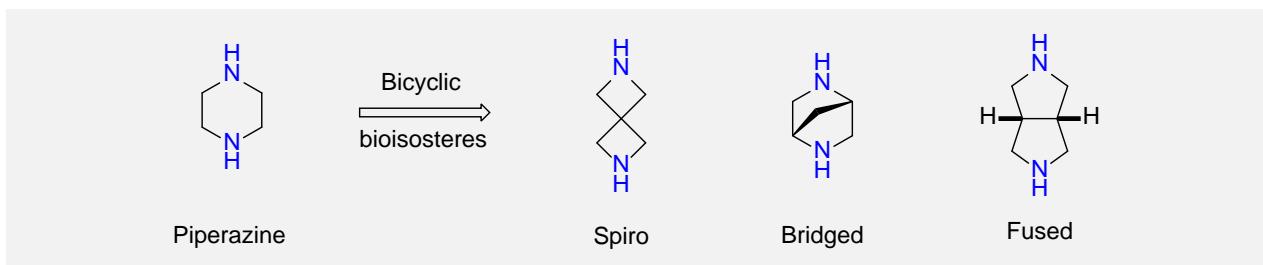
Piperazine Bioisosteres for Drug Design

Introduction

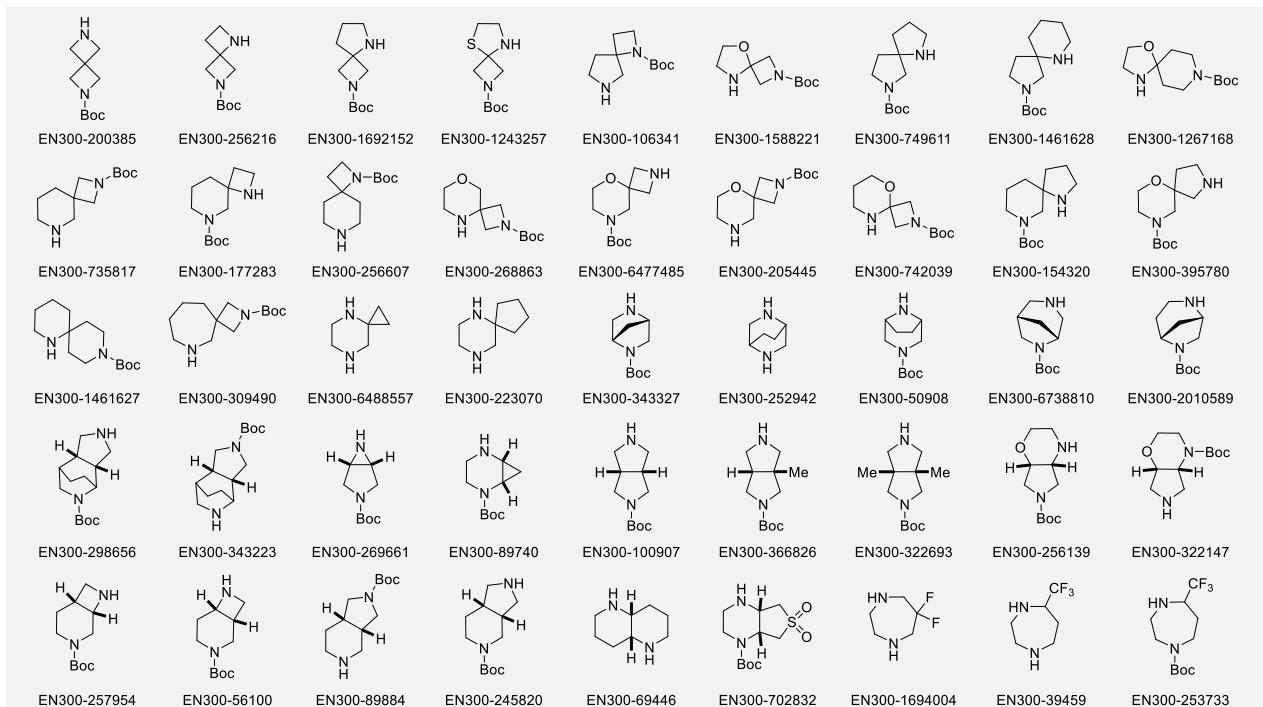
More than 100 FDA-approved drugs contain the piperazine moiety.¹ Piperazine-based analogues may advantageously alter important pharmacokinetic properties when grafted onto molecular scaffolds.²⁻⁵ In 2018, chemists showed that replacing a piperazine ring in the drug Olaparib with the spirodiamine analogue beneficially affected activity and reduced cytotoxicity of the parent compound.⁶ Herein we have designed and synthesized a library of piperazine analogues for drug design.



Design



We offer >100 unique piperazine analogues on a 5-50 g scale from stock.



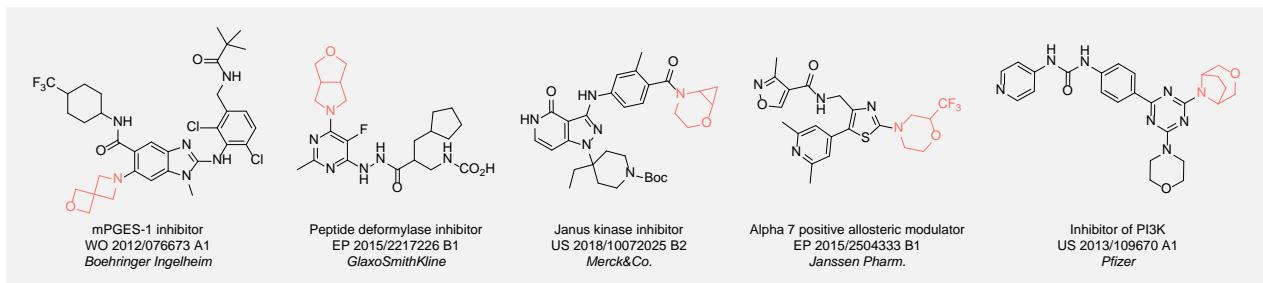
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2. J. A. Burkhard et al. *ANIE*. 2010, 49, 3524.
3. B. Chalik et al. *Chem. Eur. J.* 2017, 23, 16782.
4. B. Chalik et al. *Eur. J. Org. Chem.* 2017, 31, 4530.
5. A. Kirichok et al. *Chem. Eur. J.* 2018, 24, 5444.
6. S. W. Reilly et al. *J. Med. Chem.* 2018, 61, 5367.

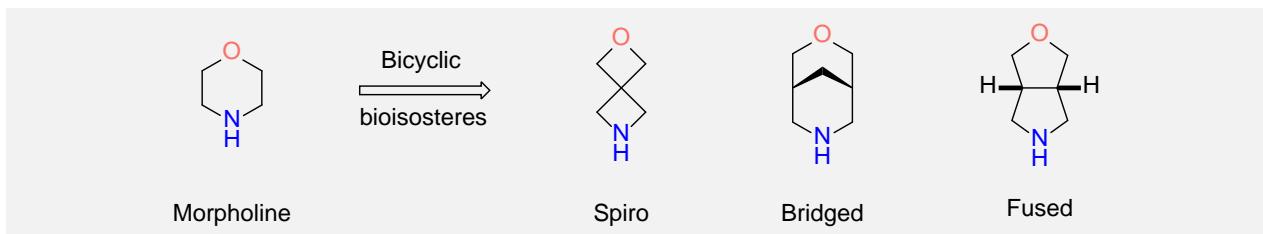
Morpholine Bioisosteres for Drug Design

Introduction

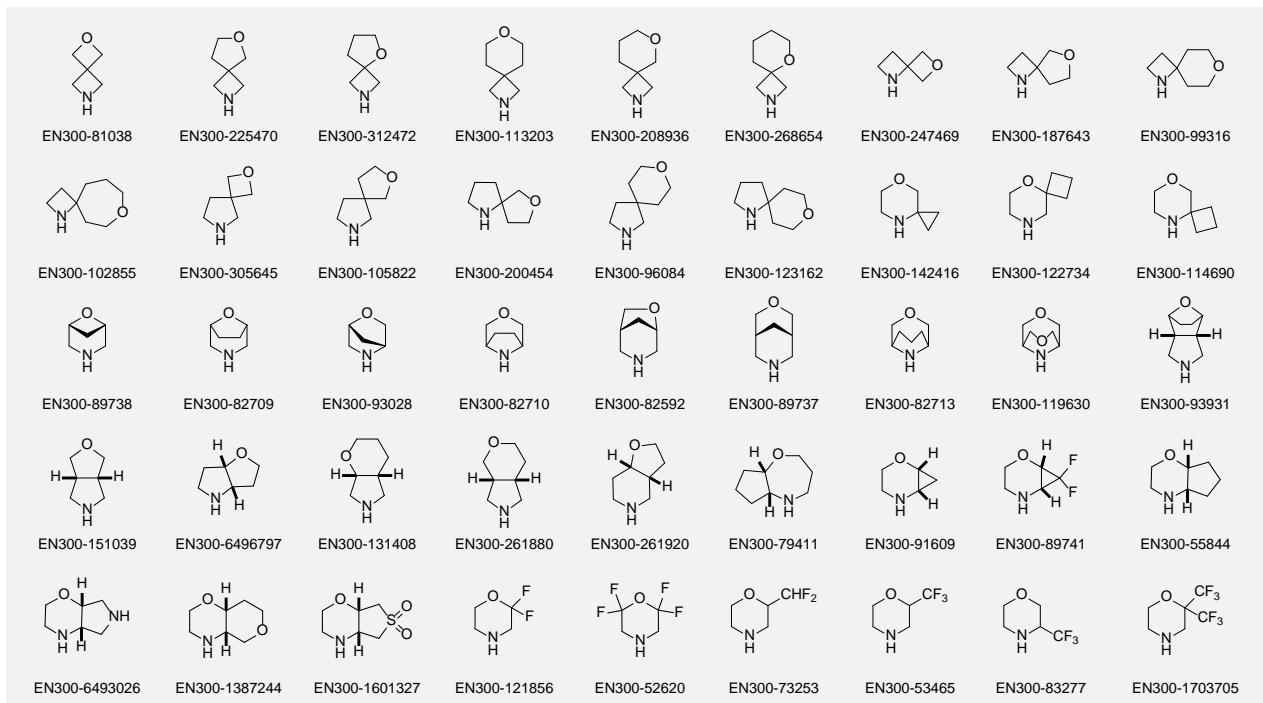
More than 20 FDA-approved drugs contain the morpholine moiety, although it is often metabolically labile.¹ Morpholine-based analogues may advantageously alter important pharmacokinetic properties such as lipophilicity and metabolic stability when grafted onto molecular scaffolds.^{2,3} Herein we have designed and synthesized a library of morpholine analogues for drug design.⁴⁻⁶



Design



We offer >100 unique morpholine analogues on a 5-50 g scale from stock.



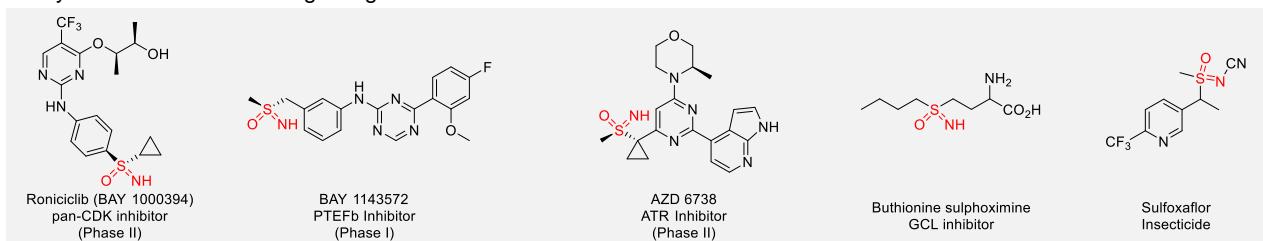
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 6. A. Kirichok et al. *Chem. Eur. J.* **2018**, 5444.

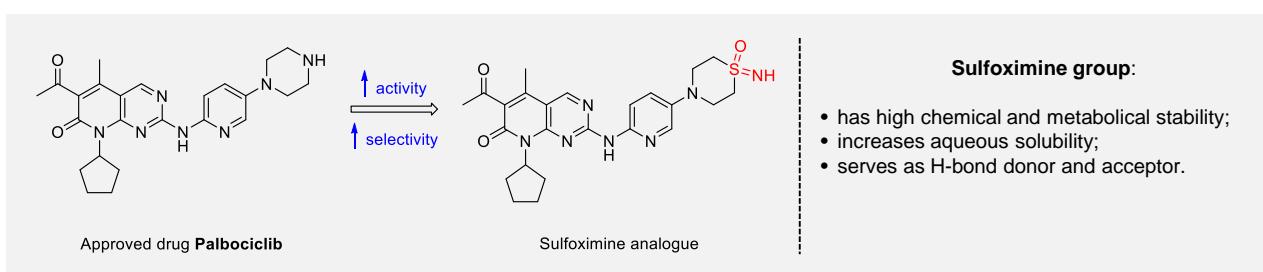
Sulfoximines for Drug Design

Introduction

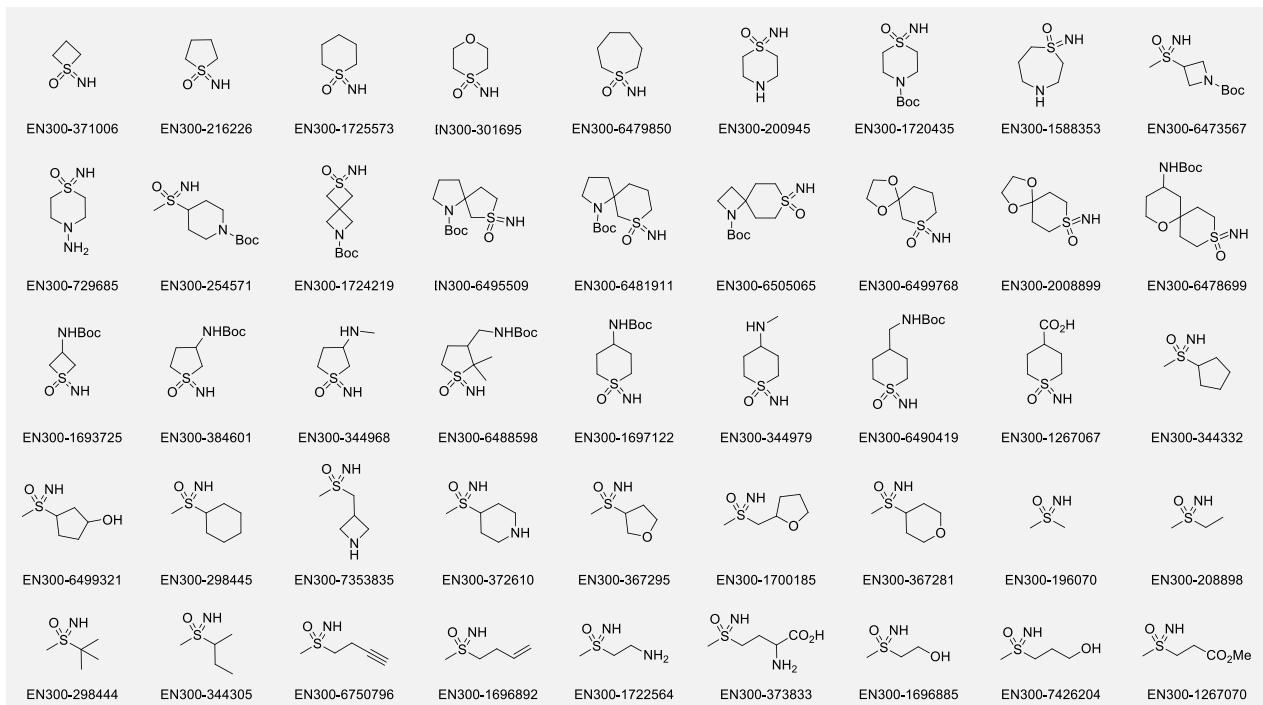
Incorporation of sulfoxime group into bioactive compounds often improves their ADME/Tox profile, and enhances potency. Moreover, the moiety of sulfoxime is chemically and metabolically stable. NH-sulfoximines can serve as the both hydrogen bond donors and acceptors at the same time.¹⁻⁶ Production and commercialization of the building blocks that already contain sulfoxime group allow significant accelerating discovery of drug candidates. Herein we have designed and synthesized a library of sulfoximines for drug design.



Design



We offer >100 unique sulfoximines on a 5-50 g scale from stock.



References

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2. M. Frings et al. *Eur. J. Med. Chem.* **2017**, 225.
3. U. Lücking. *Angew. Chem. Int. Ed.* **2013**, 9399.
4. S. Park et al. *Bioorg. Med. Chem. Lett.* **2011**, 4888.
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6. D. P. Walker et al. *Bioorg. Med. Chem. Lett.* **2009**, 3253.



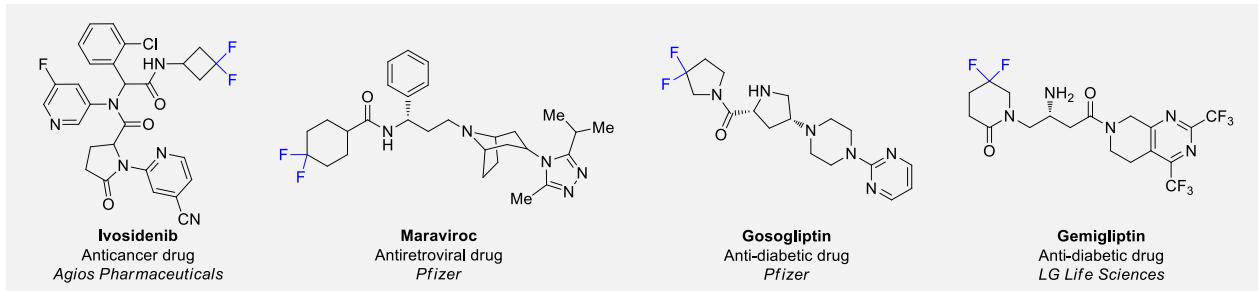
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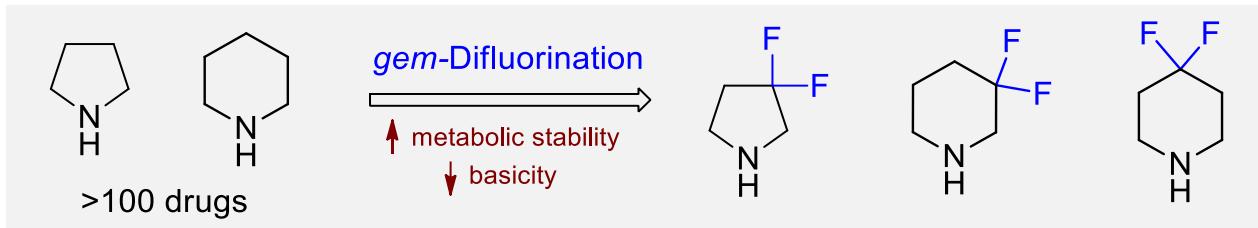
gem-Difluorinated Amines for Drug Design

Introduction

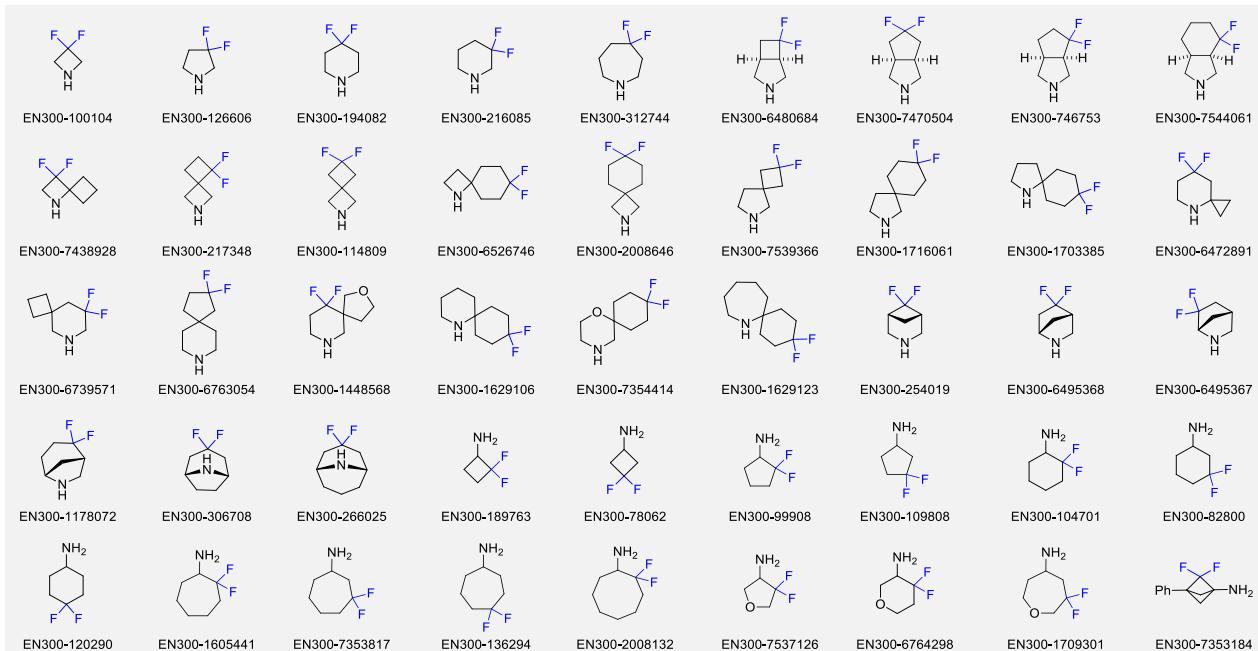
Fluorinated derivatives play an important role in medicinal chemistry. The selective incorporation of a fluoroalkyl group into bioactive compounds often affects their binding affinity, metabolic stability, lipophilicity, membrane permeability and bioactivity. *gem*-Difluoromethylene group (CF_2) is a valuable fluorinated motif that is present in pharmaceuticals and biologically active compounds. In particular, *gem*- CF_2 group improves ADME- and PK-properties.¹⁻⁶ In this context, Enamine offers a library of unique difluoro-substituted cyclic amines for drug design.



Concept



We offer: >100 *gem*-difluorinated amines on gram-scale from stock.



References

1. E. P. Gillis et al. *J. Med. Chem.* **2015**, *58*, 21, 8315.
2. X. Ma et al. *Org. Lett.* **2019**, *21*, 18, 7199.
3. R. M. Bychek et al. *J. Org. Chem.* **2019**, *84*, 23, 15106.
4. S. Purser et al. *Chem. Soc. Rev.* **2008**, *37*, 320.
5. Z. Feng et al. *Acc. Chem. Res.* **2018**, *51*, 2264.
6. Y.-L. Liu et al. *Asian J. Org. Chem.* **2013**, *2*, 194.



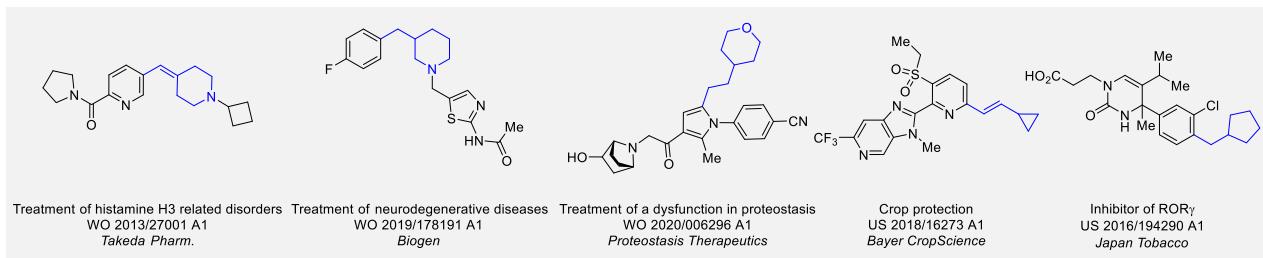
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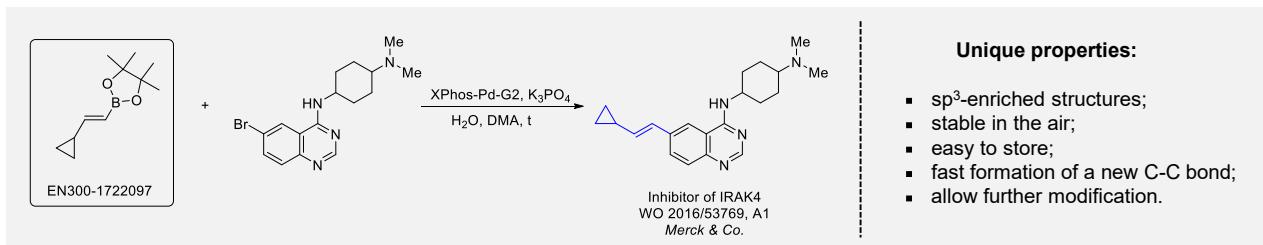
Functionalized Vinyl Boronates for C-C couplings

Introduction

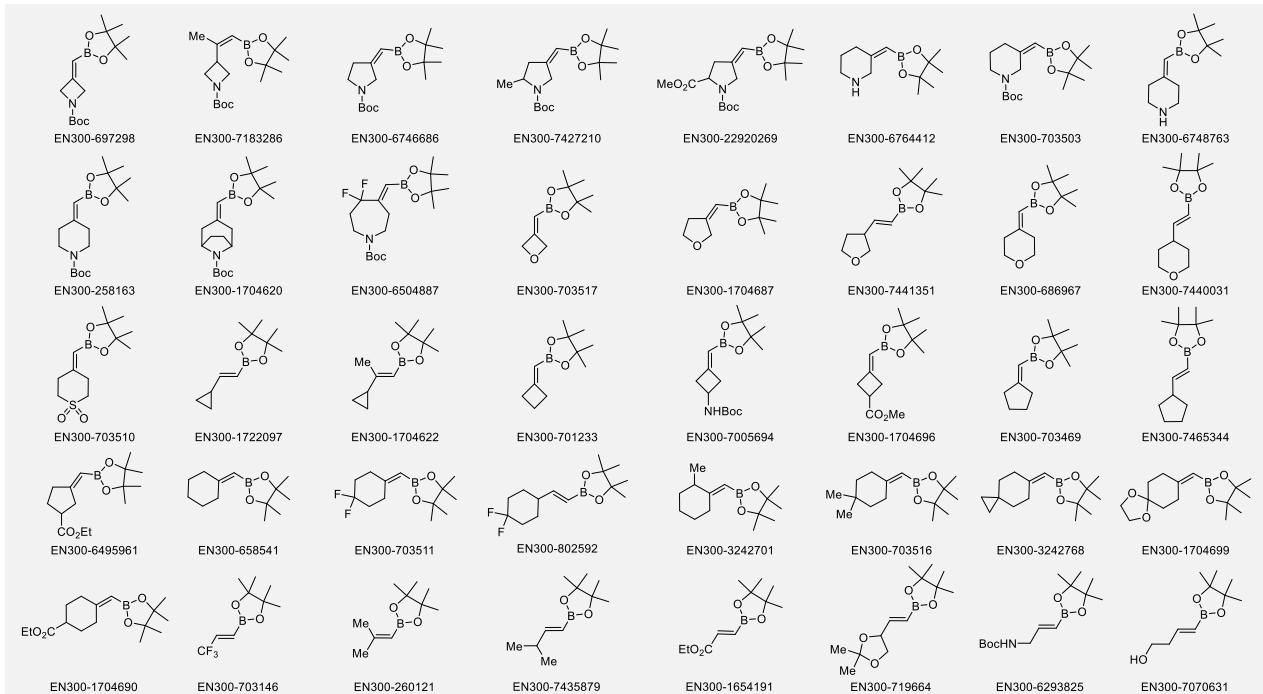
The Suzuki–Miyaura cross-coupling of boronic acid derivatives is one of the most used reactions in organic and medicinal chemist's toolbox. The rapid advancement of this method resulted in its efficient application for the late-stage modification of biologically active substrates and construction of combinatorial libraries. One of the recent trends in the field of organoboron reagents is related to the shift from aromatic compounds towards cyclic sp^3 -enriched structures, which comply with criteria of lead-oriented synthesis.^{1–6} In this context, Enamine offers a library of alkanylboronic esters for metal-mediated couplings.



Case studies



We offer: more than 50 of vinyl boronates from stock on a 5-10 g scale.



References

1. M. Kovalenko et al. *Eur. J. Org. Chem.* **2019**, 5624.
2. D. G. Brown et al. *J. Med. Chem.* **2016**, 59, 4443.
3. S. D. Roughley et al. *J. Med. Chem.* **2011**, 54, 3451.
4. P. Schäfer et al. *Nat. Commun.* **2018**, 9, 16216.
5. P. S. Campbell et al. *Chem. Commun.* **2018**, 54, 46.
6. A. Nadin et al. *Angew. Chem. Int. Ed.* **2012**, 51, 1114.



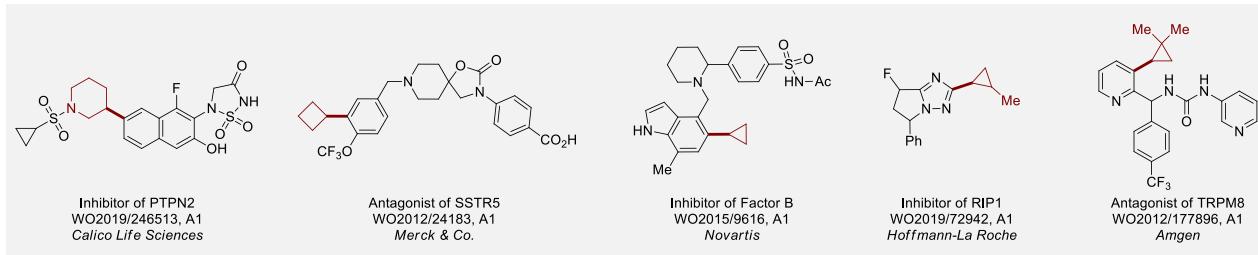
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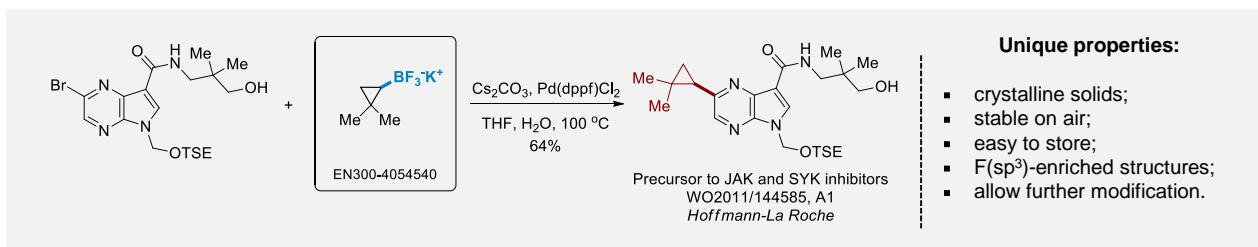
Aliphatic Trifluoroborates (-BF₃)_n for C-C couplings

Introduction

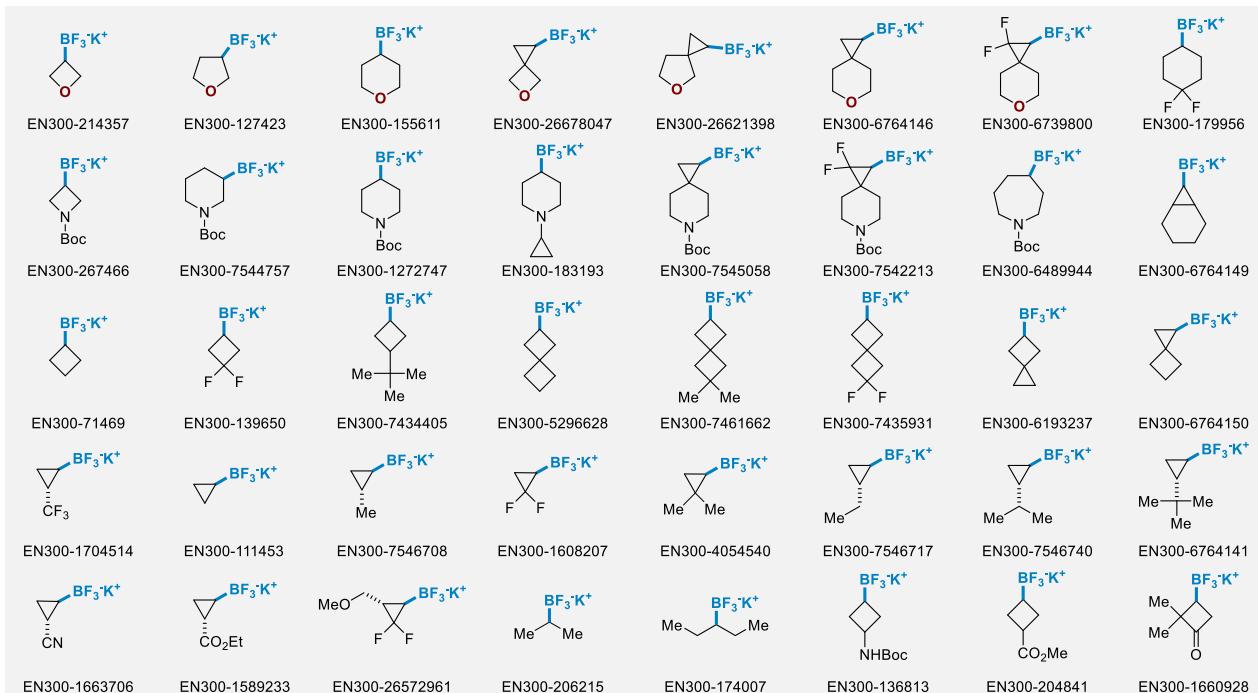
Saturated organotrifluoroborates are crystalline bench stable solids used in the standard and photoredox-accelerated Suzuki-Miyaura reaction as well as other transition-metal-catalyzed cross-couplings. One of the recent trends in the field of organoboron reagents is related to the shift from aromatic compounds towards cyclic F(sp³)-enriched structures, which comply with criteria of lead-oriented synthesis. Increasing the number of C(sp³)-hybridized carbons is a way to make a compound more drug-like.¹⁻⁶ In this context, Enamine offers a library of saturated organotrifluoroborates for metal-mediated couplings.



Case studies



We offer: >50 of aliphatic trifluoroborates from stock on a 5-10 g scale.



References

- A. W. Dombrowski et al. *ACS Med. Chem. Lett.* **2020**, 11, 597.
- G. A. Molander et al. *Curr. Opin. Drug Discov. Devel.* **2009**, 811.
- O. Hryschuk et al. *Eur. J. Org. Chem.* **2020**, 2217.
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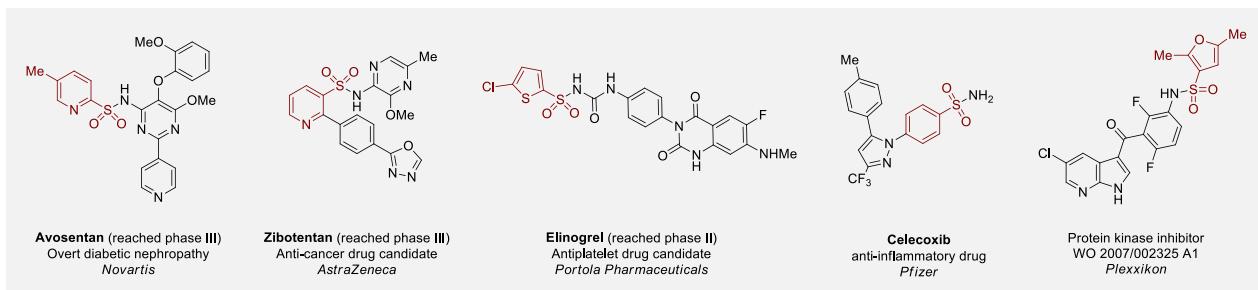
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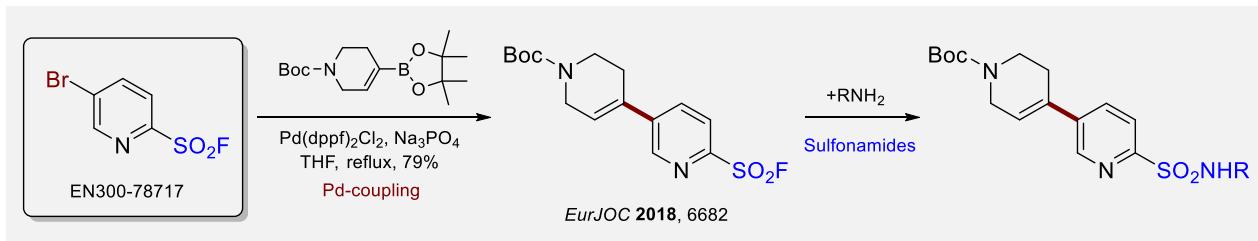
Heterocyclic Sulfonyl Fluorides for Pd-Catalyzed C–C Coupling Reactions

Introduction

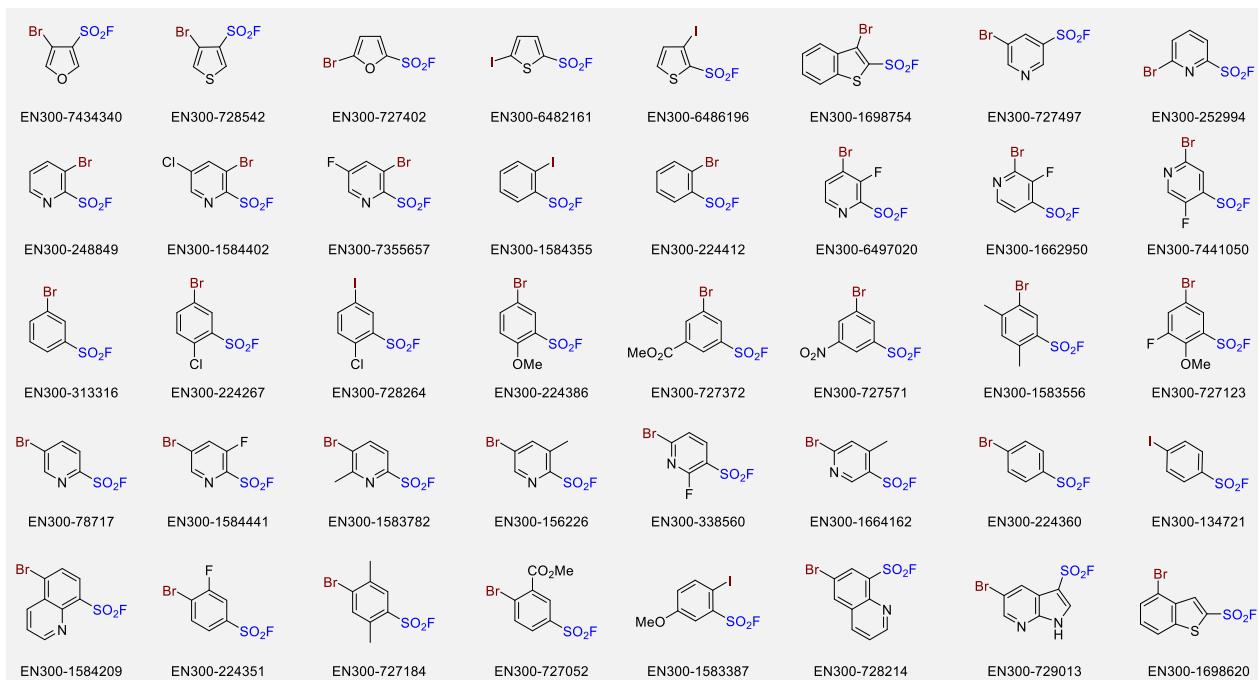
Hetaryl bromides and iodides bearing SO_2F group are versatile substrates for metal catalyzed crosscoupling reactions. Due to the high energy of the S–F bond, sulfonyl fluorides are stable toward hydrolysis, metal catalysis, or reductive reaction conditions. On the other hand, they undergo selective nucleophilic substitution at the sulfur(VI) electrophilic center under controllable reaction conditions.^{1–5} Herein we have designed and synthesized a library of bifunctional building blocks for drug design.



Design



We offer



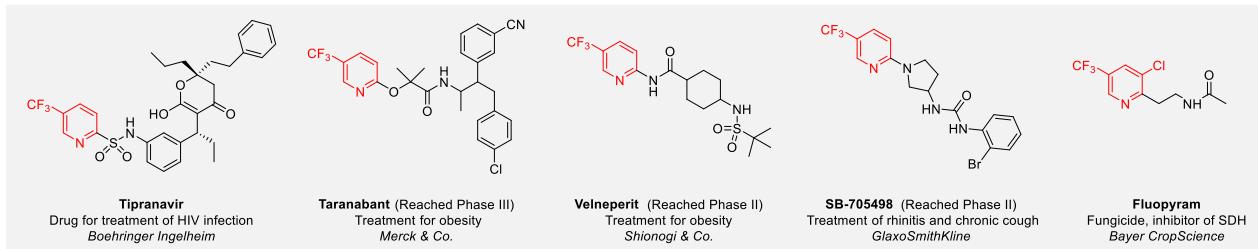
References

- A. Y. Cherepkha et al. *Eur. J. Org. Chem.* 2018, 47, 6682.
- Y. L. Hsu et al. *Tetrahedron* 2016, 72, 58.
- S. A. Zhersh et al. *Chem. Eur. J.* 2018, 24, 8343.
- P. K. Chinthakindi et al. *Eur. J. Org. Chem.* 2018, 3648.
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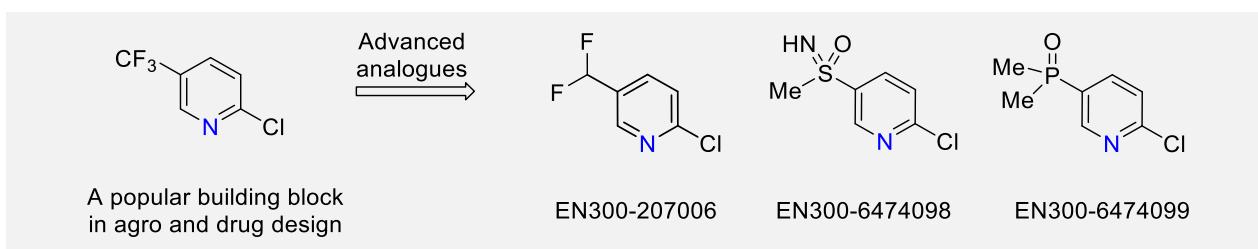
Analogues of CF₃-Pyridine for Drug Design

Introduction

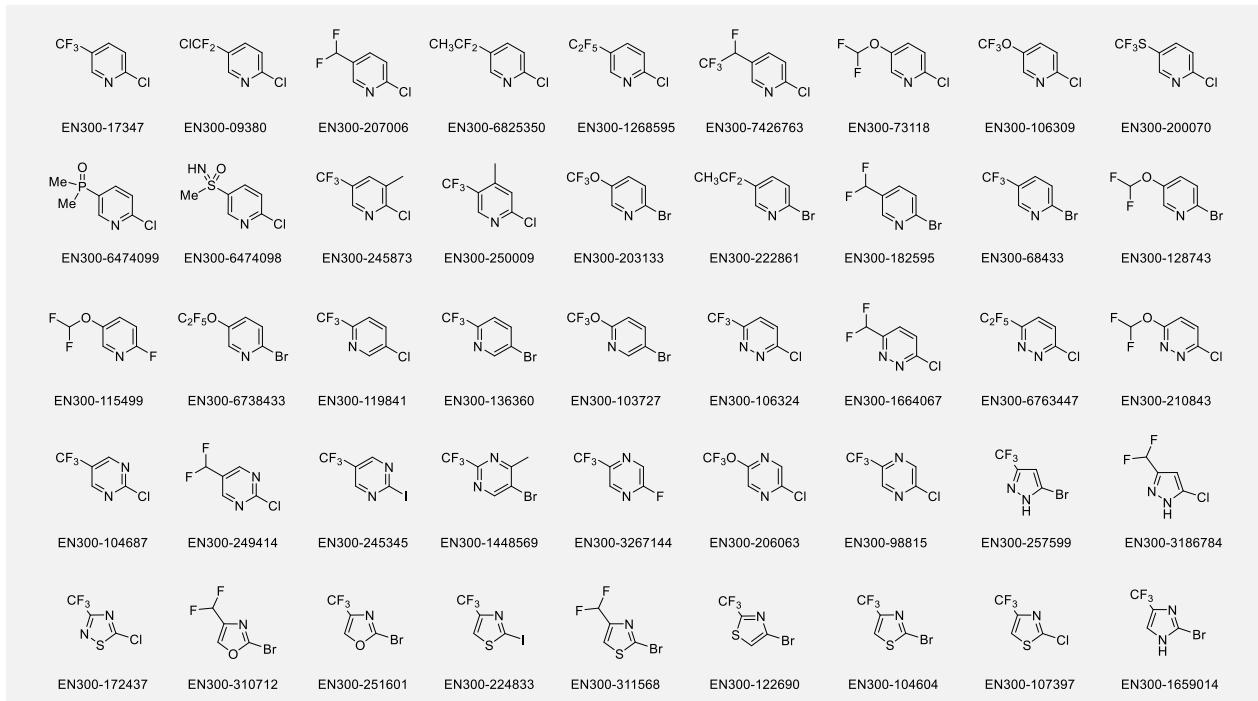
More than 100 FDA-approved drugs contain pyridine ring and another 100 have been in clinical trials.¹ Examples containing CF₃-pyridine moiety include the antiviral drug *Tipranavir* and drug candidates *Taranabant*, *Velneperit*, *SB-705498*. Besides, CF₃-pyridine containing compounds have been playing a fundamental role in agrochemistry.^{2,3} The introduction of fluorine-containing group alters important pharmacokinetic properties of molecular scaffolds.⁴ Herein we have designed and synthesized a library of CF₃-substituted pyridine analogues for drug design and agrochemistry.



Design



We offer >100 unique CF₃-pyridine analogues on a 5-50 g scale from stock



References

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2. A.-Y. Guan et al. *Bioorg. Med. Chem.* **2016**, *24*, 342.
3. *Bioactive Heterocyclic Compound Classes: Agrochemicals*, Wiley-VCH, **2012**, 209-223.
4. *Fluorine in Life Sciences: Pharmaceuticals, Medicinal Diagnostics, and Agrochemicals*, Academic Press, **2019**, 181-211.



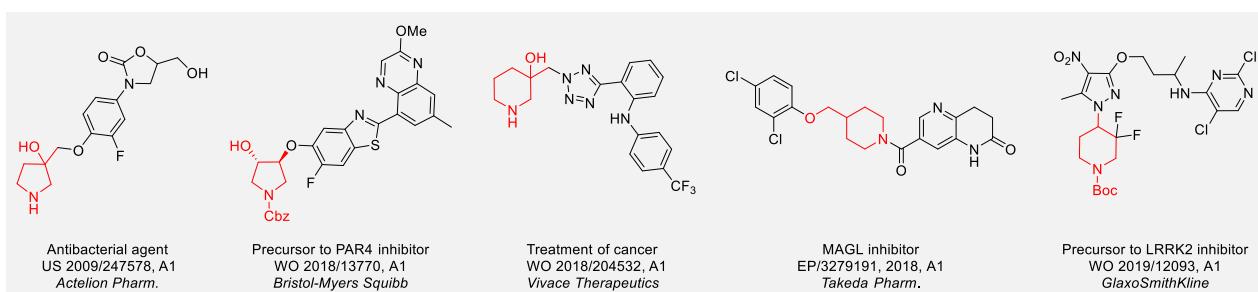
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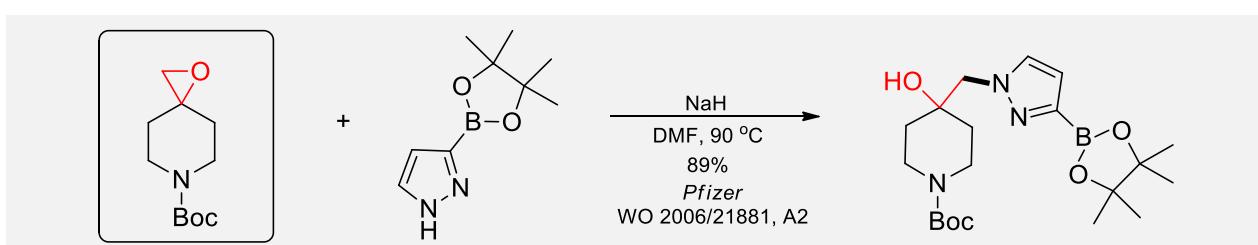
Epoxides for Drug Design

Introduction

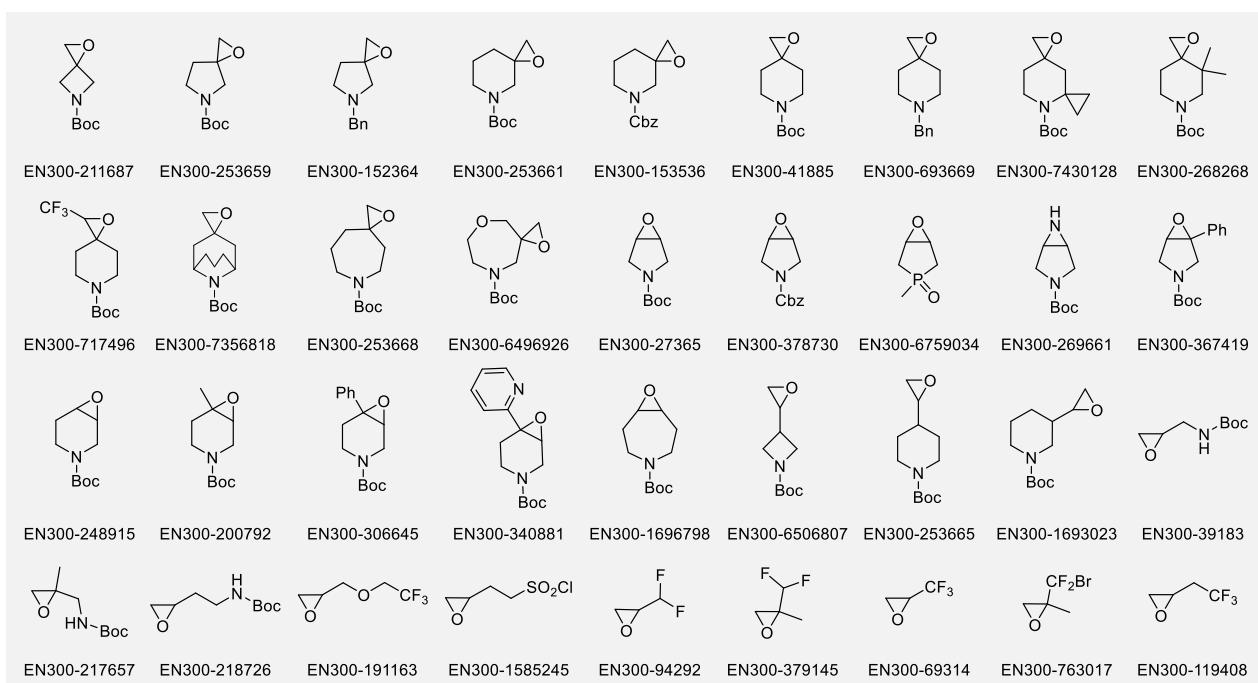
Epoxides are important heterocyclic units found in various naturally occurring molecules, some with potential bioactivities. At least fourteen drugs on the market are epoxide-containing.¹ In addition, epoxides are valuable building blocks in medicinal chemistry. They react with nucleophiles in a ring-opening process to form new C-C, C-O and C-N bonds.²⁻⁵ Herein we have designed and synthesized a library of small heteroaliphatic epoxides for drug design.



Design



We offer >100 unique epoxides on a 5-50 g scale from stock.



References

1. www.ebi.ac.uk/chembl/.
2. E. N. Jacobsen, *Acc. Chem. Res.* **2000**, *33*, 421.
3. T. B. Hughes et al. *ACS Cent Sci.* **2015**, *4*, 168.
4. D. M. Hodgson et al. *Tetrahedron*, **1996**, *52*, 14361.
5. G. Dake. *Comprehensive Heterocyclic Chemistry III, 1.03 Oxiranes and Oxirenes: Monocyclic*, **2008**, p. 173.



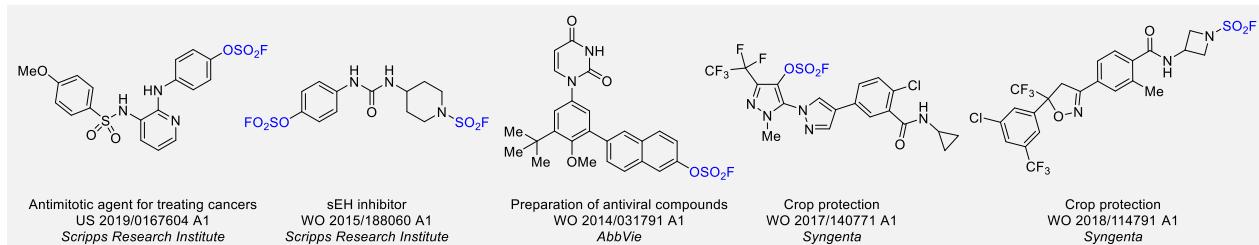
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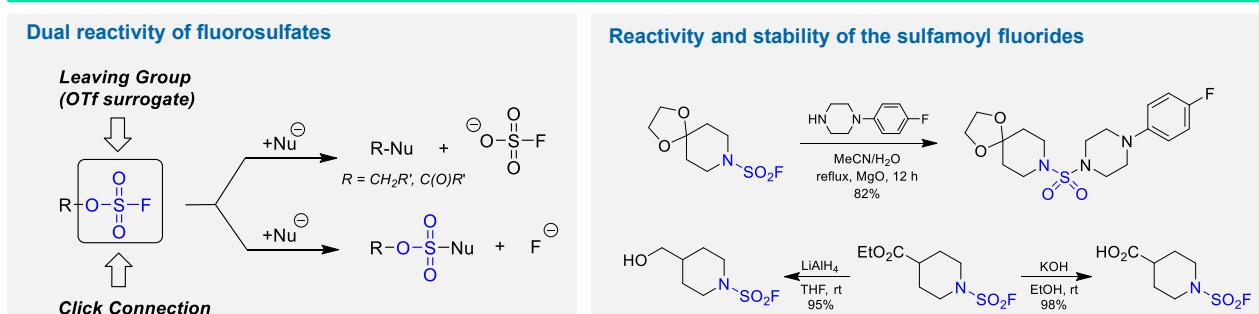
Fluorosulfates and Sulfamoyl Fluorides for Drug Design

Introduction

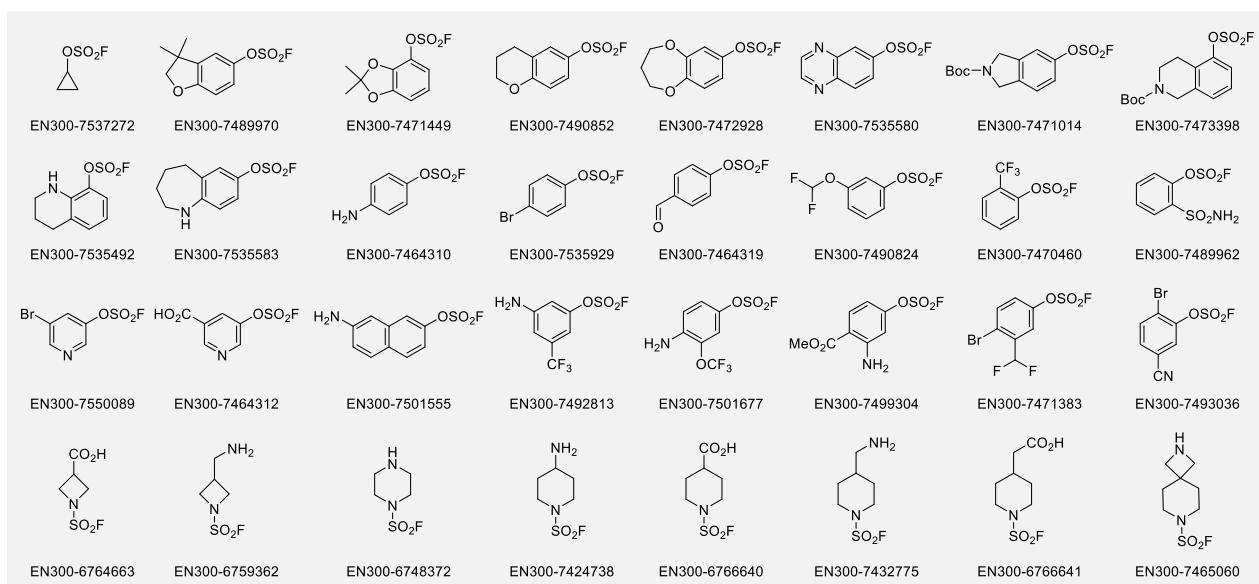
Aryl fluorosulfates and sulfamoyl fluorides are widely used in chemical biology, medicinal chemistry and agrochemistry. The former exhibit chemoselective reactivity with the side chains of tyrosine, lysine, serine and histidine in the proteins, and can be used to target non-enzymes as well as enzymes. Depending on the nature of the substituent, the $-\text{OSO}_2\text{F}$ unit can be a good leaving group or a robust connector. The fluorosulfates are quite stable toward hydrolysis under neutral or acidic conditions. The *N*-disubstituted sulfamoyl fluorides are stable toward hydrolysis under basic condition, inert toward a wide range of nucleophiles and dramatically more robust than analogous chlorides.¹⁻⁶ In this context, Enamine offers a library of unique aryl fluorosulfates and sulfamoyl fluorides for drug design.



Properties of ROSO_2F and $\text{R}_2\text{NSO}_2\text{F}$



We offer: >100 unique fluorosulfates and sulfamoyl fluorides in gram amounts in stock.



References

1. J. Dong et al. *Angew. Chem. Int. Ed.* **2014**, *53*, 9430.
2. H. Zhou et al. *Org. Lett.* **2018**, *20*, 812.
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4. W. Chen et al. *J. Am. Chem. Soc.* **2016**, *138*, 7353.
5. L. H. Jones. *ACS Med Chem Lett.* **2018**, *9*, 584.
6. W. Chen et al. *Angew. Chem. Int. Ed.* **2016**, *55*, 1835.



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Silicon-Containing Building Blocks for Drug Design

Introduction

Silicon-containing compounds have been largely ignored in drug design until recently.¹ Silicon can be considered a bioisostere of carbon and hence offers an innovative avenue in drug discovery. For example, C/Si exchange in drug-like scaffolds provides an exciting approach in medicinal chemistry to improve ADME/Tox profile and to enhance potency of the biologically active compounds (Figure 1).²⁻⁶ Herein we have designed and synthesized a library of silicon-containing building blocks for drug design.

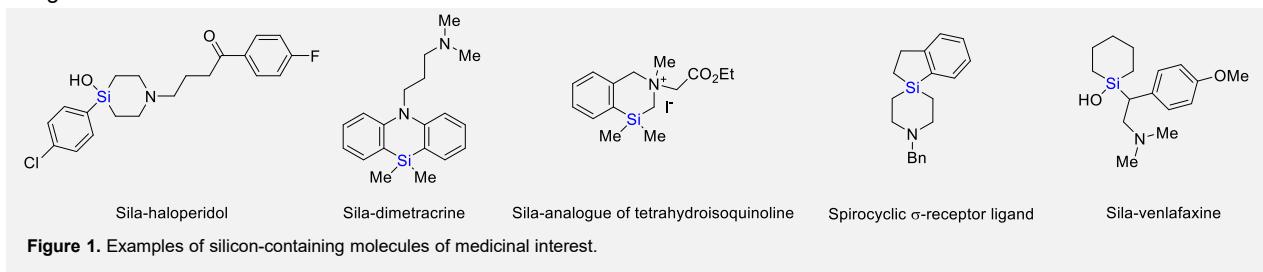
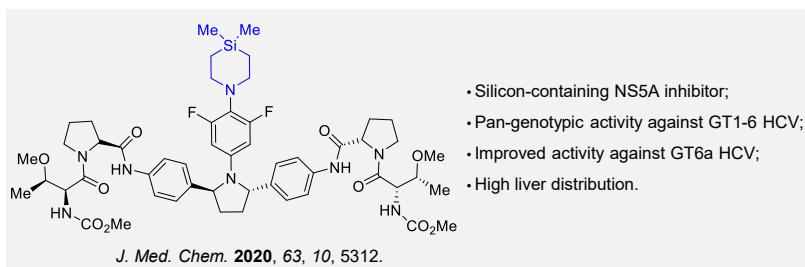


Figure 1. Examples of silicon-containing molecules of medicinal interest.

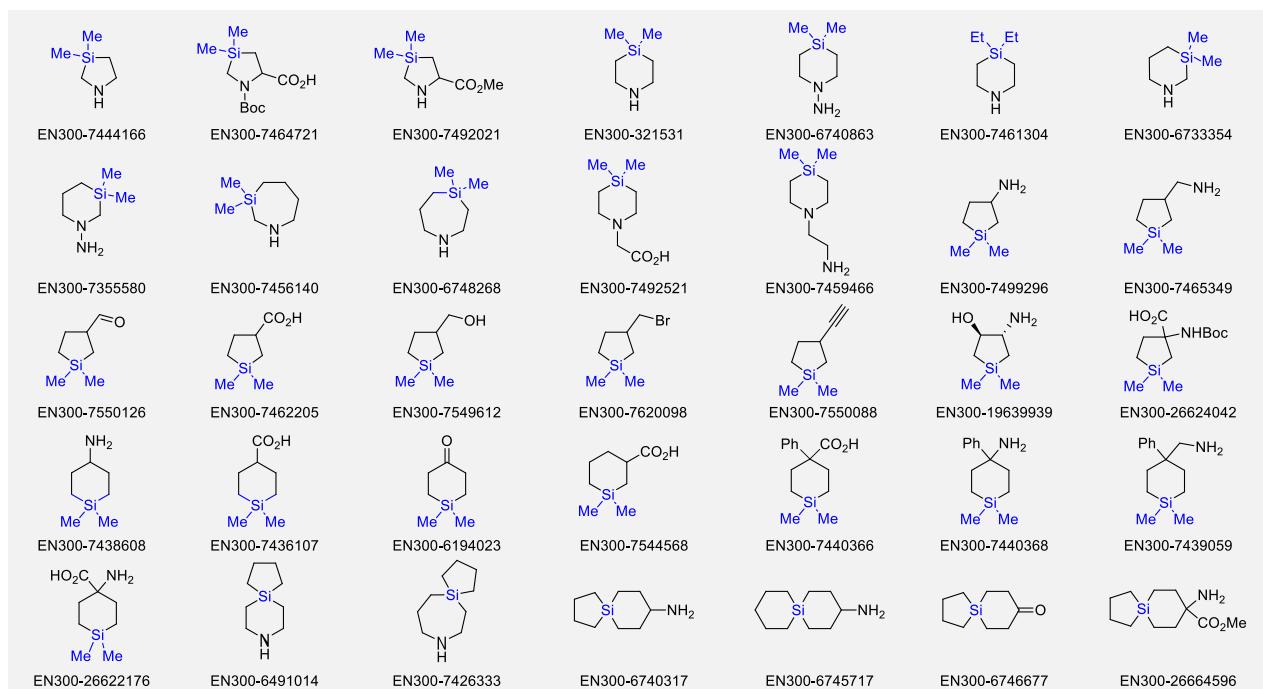
Design



Advantages of “Silicon Switch”:

- Bond length C–C = 1.54 Å, C–Si = 1.87 Å → changes in the interactions with specific proteins;
- Increase in lipophilicity;
- Si prefers higher coordination numbers → access to compounds for which corresponding carbon analogs are not available.

We offer more than 30 of silicon-containing building blocks from stock on a 5-10 g scale.



References

1. S. J. Barraza et al. *J. Am. Chem. Soc.* **2018**, *140*, 6668.
2. R. Ramesh et al. *J. Med. Chem.* **2018**, *61*, 3779.
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5. A. K. Franz et al. *J. Med. Chem.* **2013**, *56*, 388.
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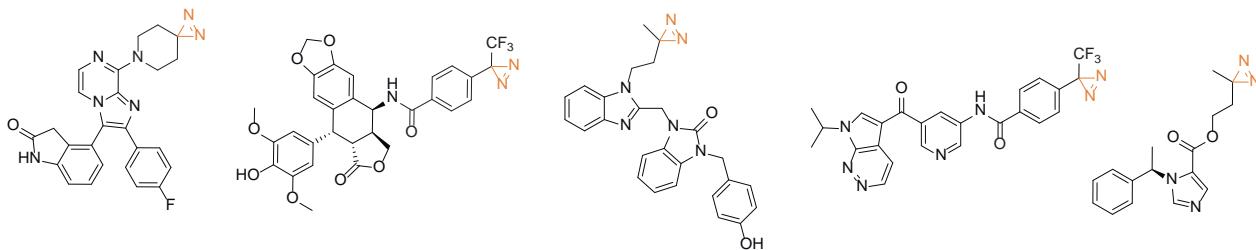


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DIAZIRINES

Diazirine is a smallest heterocycle that is stable in the dark, but forms reactive carbene upon irradiation with light. Its introduction into the structures of biologically active compounds accompanying with only minor change in MW (plus only 2 nitrogen atoms!) has proven to provide efficient tools to study interactions with biological targets including their isolation and identification. Given the success and progress in the field of activity-based protein profiling, the use of diazirine photolabeling will most likely continue to rise and it is important to have a commercial access to diverse diazirine-containing building blocks.



Antagonist of GRIA1
WO 2016/176457
Janssen Pharm.

Analog of etoposide
Bioorg. Med. Chem.
2010, 830.

Inhibitor of RSV
Bioorg. Med. Chem.
2004, 1133.

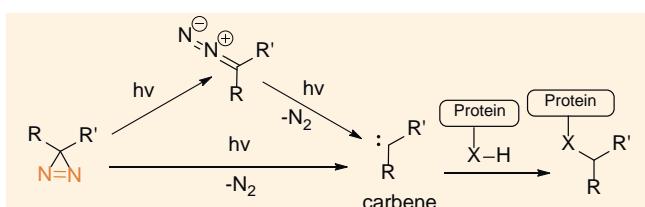
Inhibitor of TrkA
US 2012/258950
Pfizer

Blocker of nAChRs
Mol. Pharmacol.
2009, 1084.

Properties

- smallest photoreactive group
- excitation at 355 nm
- high chemical stability

Upon irradiation of a ligand-target complex, a diazirine-containing ligand generates a reactive carbene that covalently binds the ligand to the target.



Our offer: >30 building blocks from stock.

Custom synthesis of the diazirine building blocks and diazirine-containing ligands.

EN300-224745	EN300-223387	EN300-226428	EN300-260811	EN300-97492
EN300-138815	EN300-84981	EN300-300644	EN300-370261	EN300-365848
EN300-222270	EN300-222268	EN300-223410	EN300-311025	EN300-315165

References

- ¹ L. Dubinsky *et al.* *Bioorg. Med. Chem.* **2012**, 554.
² N. Burkard *et al.* *Eur. J. Org. Chem.* **2010**, 2176.

- ³ A. Blencowe *et al.* *Soft Matter.* **2005**, 178.
⁴ Hatanaka *et al.* *Curr. Top. Med. Chem.* **2002**, 271.



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Structurally optimized tetrazines for rapid biological labeling

Introduction

Bioorthogonal chemical reactions are closely associated with the characteristics of “click” chemistry, occurring with high selectivity and fast reaction kinetics *in vivo*.^{1,2} Consequently, these reactions found use as multipurpose tools for chemical biology. The Inverse-electron-Demand Diels–Alder (**iEDDA**) reaction between tetrazines and strained alkenes is fairly new ligation reaction, which displays **rates 3-7 orders of magnitude faster** than many bioorthogonal reactions.³ High reaction rates, biocompatibility, together with the ability of tetrazines to quench fluorescence of some fluorophores, widely used for fluorescent labeling, and recover it after **iEDDA** reaction (**Figure 1**) make tetrazine derivatives unique and versatile tools for bioorthogonal chemistry. **Figure 2** is showcasing possible approach to modification of commonly used fluorophore as fluoresceine (**A**) with tetrazines⁴ and application of tetrazine derivatives in DNA encoded libraries technologies (DELT), as the core scaffolds (**B**).⁵

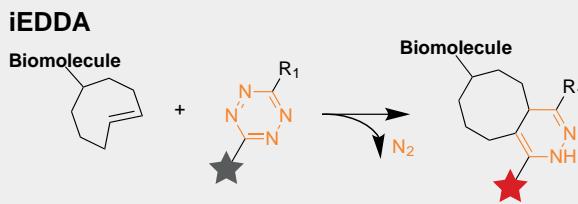


Figure 1. General scheme of **iEDDA** ligation.

Application

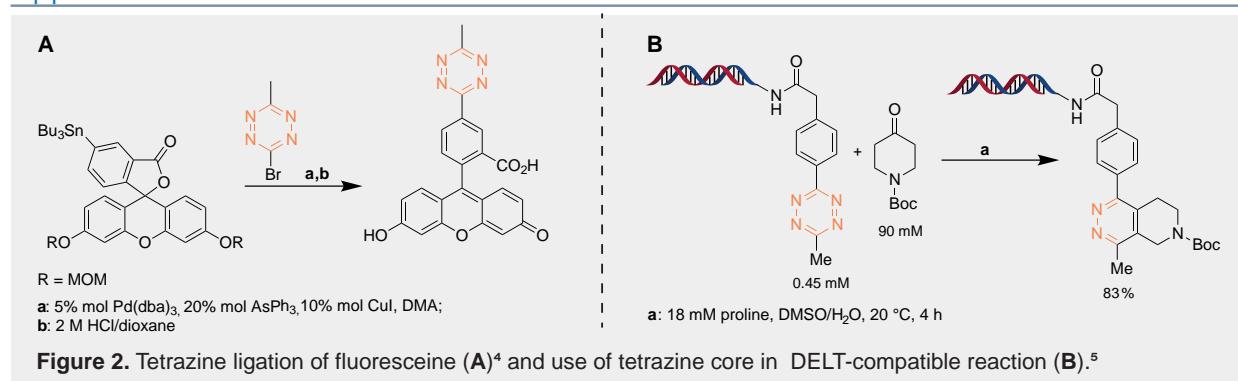
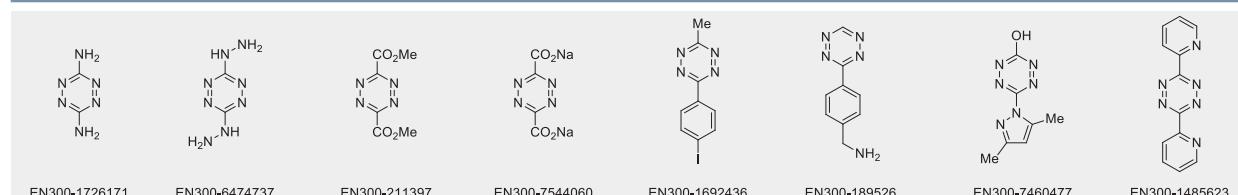
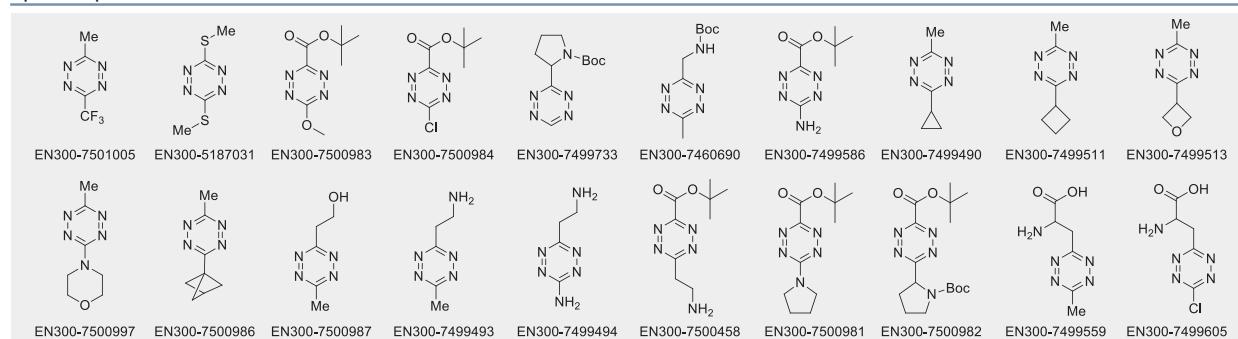


Figure 2. Tetrazine ligation of fluoresceine (**A**)⁴ and use of tetrazine core in DELT-compatible reaction (**B**).⁵

We offer Currently, we have synthesized 8 tetrazine-containing building blocks, that are available in our store on a gram scale.



Pre-order We also have designed a library of tetrazine-containing building blocks. These molecules can be synthesized upon request.



References

1. L. Carroll et al. *Org. Biomol. Chem.* **2013**, *11*, 5772.
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3. H. L. Evans et al. *Chem. Commun.* **2014**, *50*, 9557.
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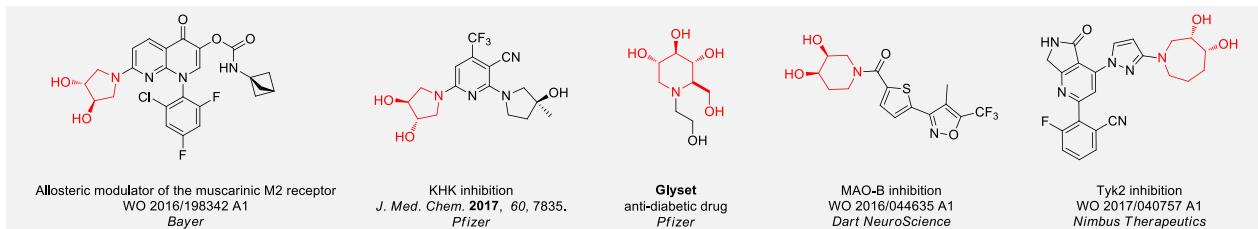
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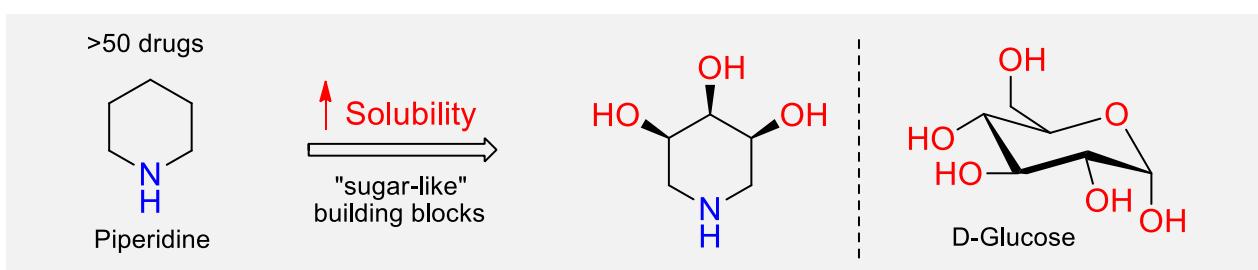
Sugar-like Building Blocks for Drug Design

Introduction

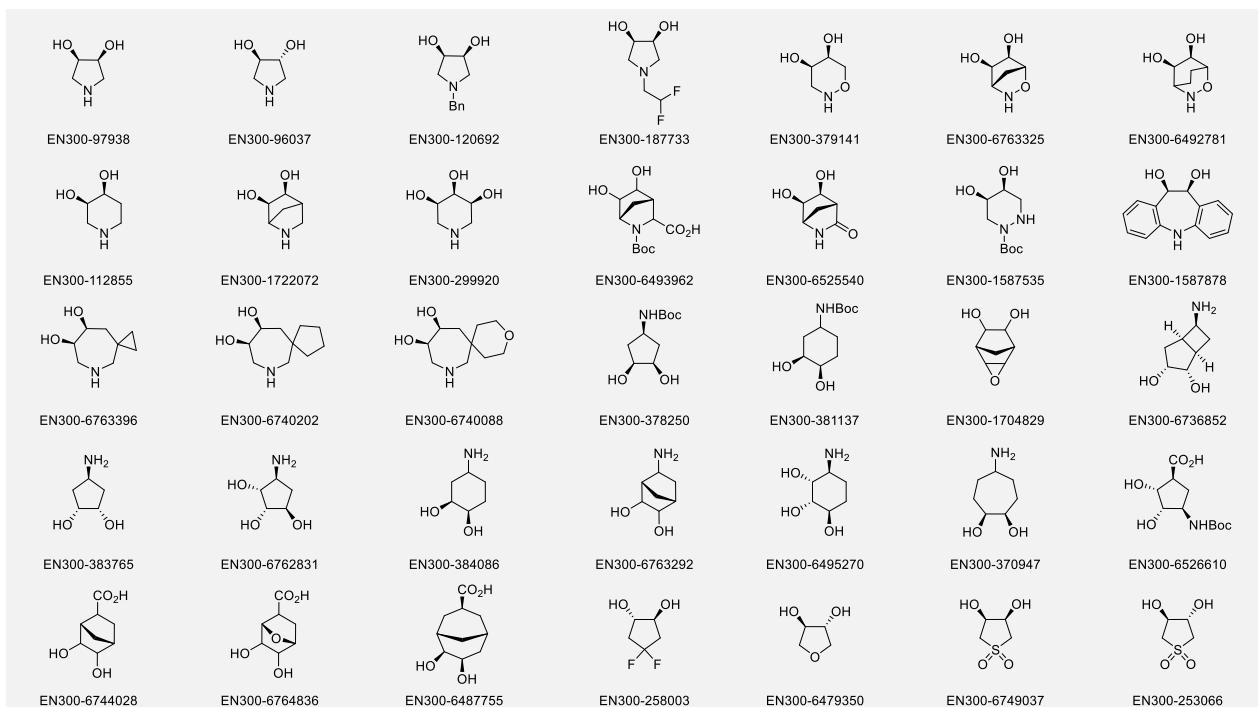
Saturated azaheterocycles are popular in modern drug discovery programs. More than 50 FDA-approved drugs containing a fragment of pyrrolidine and piperidine have appeared on the market.¹ Pyrrolidine/piperidine-based azasugars and their analogues are potent glycosidase inhibitors. These unique molecules promise a new generation of iminosugar-based medicines for a wide range of diseases.²⁻⁵ For example, a bioactive azasugar includes the anti-diabetic drug *Glyset*. In this context, herein we have designed and synthesized a library of sugar-like derivatives for drug design.



Design



We offer >50 unique sugar-like derivatives on a 5-50 g scale in stock.



References

- www.drugbank.ca
- N. F. Bras et al. *Expert Opin. Ther. Patents* **2004**, 24, 857.
- V. R. Dodd et al. *Eur. J. Org. Chem.* **2007**, 5583.
- Y. L. Merrer et al. *Bioorg. Med. Chem.* **1997**, 5, 519.
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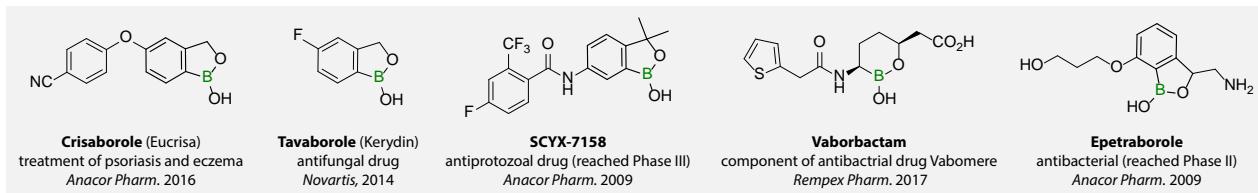
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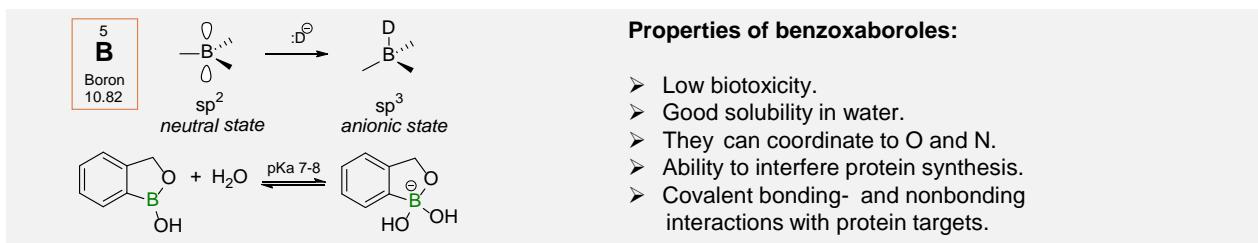
Benzoxaboroles for Drug Design

Introduction

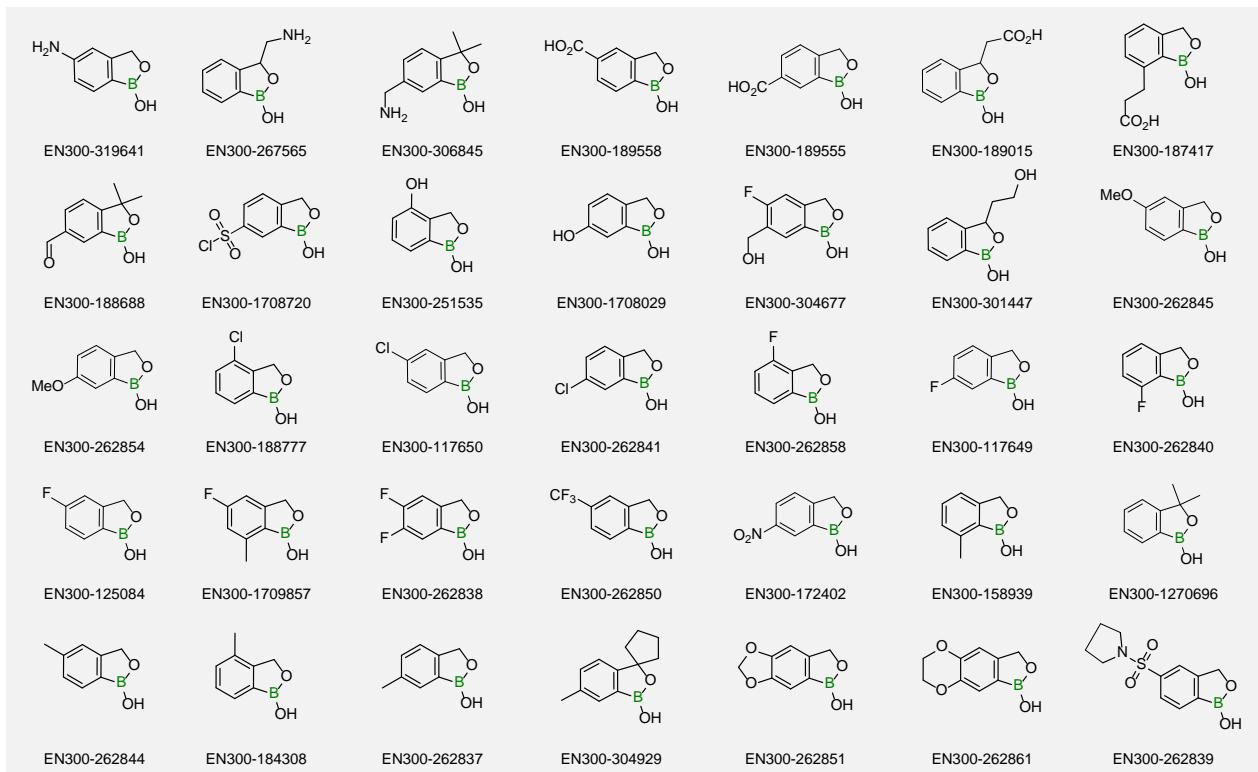
Benzoxaborole is a versatile boron-heterocyclic scaffold which has found in the last 10 years a broad spectrum of applications in medicinal chemistry. The use of benzoxaborole moiety in the design of compounds led to the discovery of new classes of anti-bacterial, anti-fungal, antiprotozoal, anti-viral and as anti-inflammatory agents with interesting drug development perspectives. Two benzoxaborole derivatives are already clinically used for the treatment of onychomycosis (*Tavaborole*) and atopic dermatitis (*Crisaborole*), with several others in various phases of clinical trials¹⁻⁹.



Advantages



We offer



References

1. S. J. Baker et al. *Chem. Soc. Rev.* **2011**, 4279.
2. F. L. Rock et al. *Science* **2007**, 1759.
3. T. Akama et al. *Bioorg. Med. Chem. Lett.* **2009**, 2129.
4. X. Li et al. *Bioorganic Med. Chem. Lett.* **2010**, 3550.
5. V. Hernandez et al. *Antimicrob. Agents Chemother.* **2013**, 1394.
6. D. B. Diaz et al. *Nat. Chem.* **2017**, 731.
7. V. M. Dembinsky et al. *Chem. Rev.* **2011**, 209.
8. A. Nocentini et al. *Expert Opin. Ther. Pat.* **2018**, 493.
9. Y. K. Zhang et al. *Bioorg. Med. Chem. Lett.* **2011**, 644.

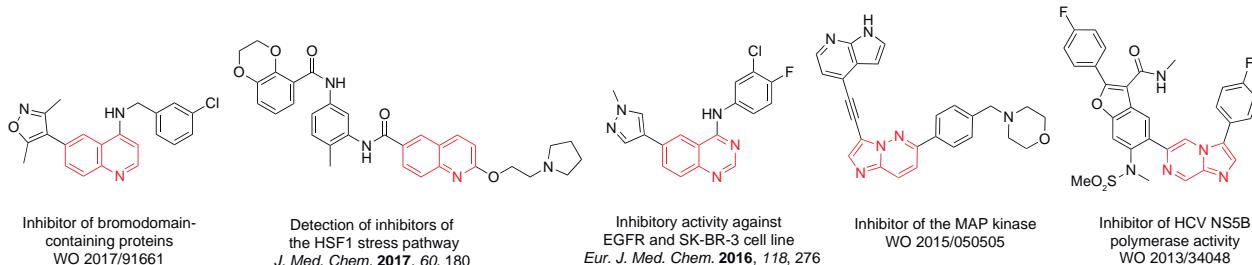


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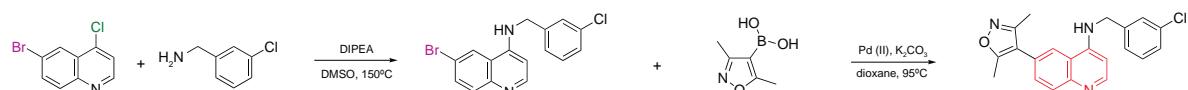
Heterocyclic scaffolds

Unsaturated heterocycles are popular in bioactive compounds and drugs. Herein, we offer a library of heterocycles with two halogen atoms bearing different activity. They can be used for the stepwise nucleophilic substitution followed by a metal-mediated cross-coupling to produce the functionalized products.

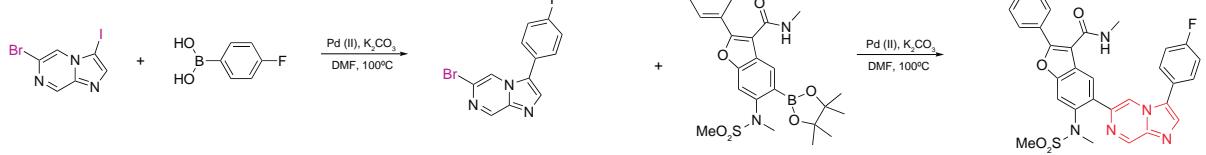


Synthesis of bioactive compounds

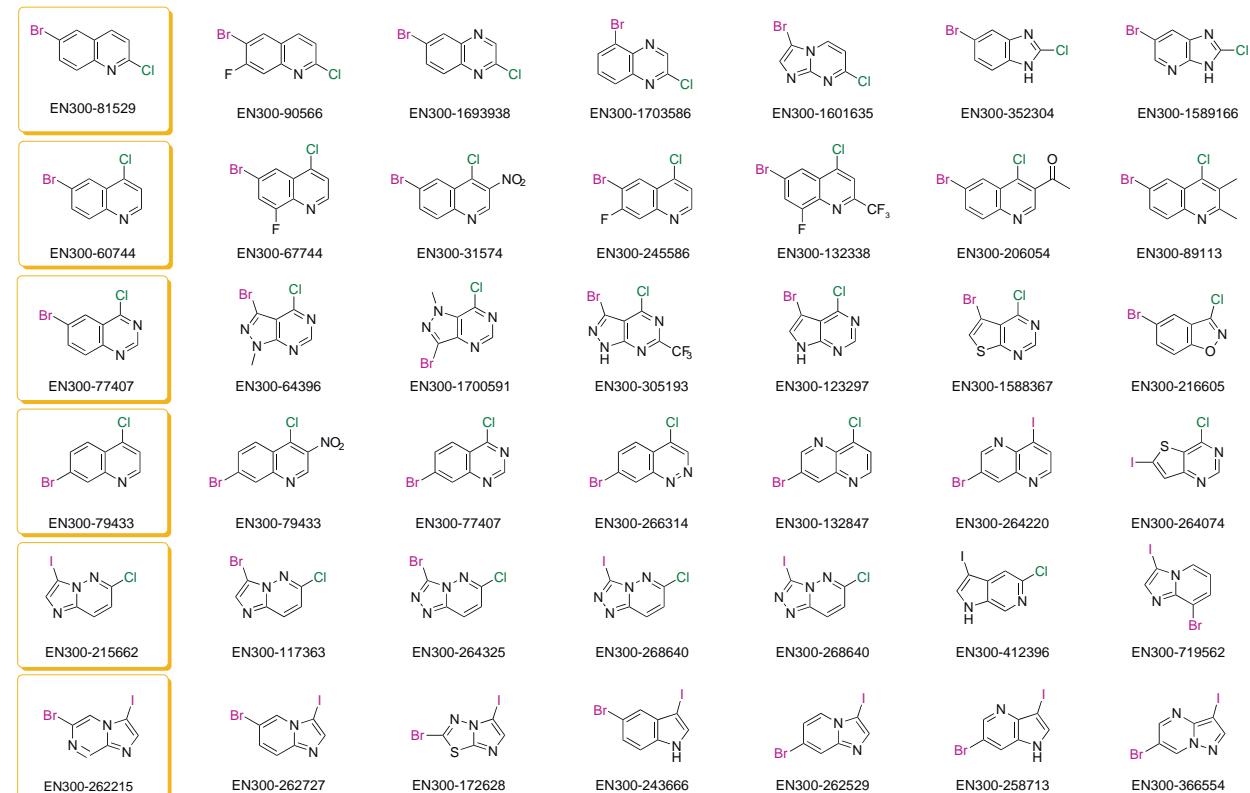
Nucleophilic substitution and Pd-mediated coupling:



Two subsequent Pd-mediated couplings:

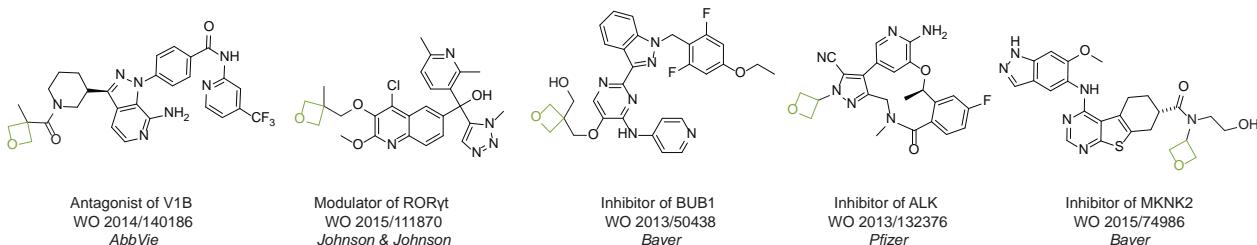


Our offer: over 100 different bicyclic heterocycles with two different halogens on a gram scale in stock.



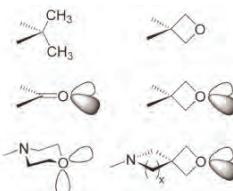
OXETANES

For more than 130 years since the first preparation by Reboul, oxetanes have largely remained neglected in medicinal chemistry. The unit of oxetane can trigger profound changes in aqueous solubility, lipophilicity, metabolic stability, and conformational preference when replacing the commonly employed functionalities such as gem-dimethyl or carbonyl groups. Of particular interest are the oxetanes substituted at the 3-position, since they remain none-chiral. At the moment, oxetane-containing building blocks flourish in medicinal chemistry and drug discovery.



Properties of Oxetanes

- high chemical stability;
- high aqueous solubility;
- low lipophilicity;
- high metabolic stability;
- hydrogen-bond acceptor ability.



Application of Oxetanes

- less lipophilic and more metabolically stable than a gem-dimethyl group;
- replacement for a metabolically and chemically labile carbonyl group;
- metabolically-robust analogue of morpholine.

Our offer: >200 oxetane-containing building blocks on gram scale in stock.

References

- ¹ G. Wuitschik *et al.* *J. Med. Chem.* **2010**, 3227.
² J.A. Burkhard *et al.* *Angew. Chem. Int. Ed.* **2010**, 9052.

- ³ G. Wuitschik *et al.* *Angew. Chem. Int. Ed.* **2008**, 4512.
⁴ G. Wuitschik *et al.* *Angew. Chem. Int. Ed.* **2006**, 7736.



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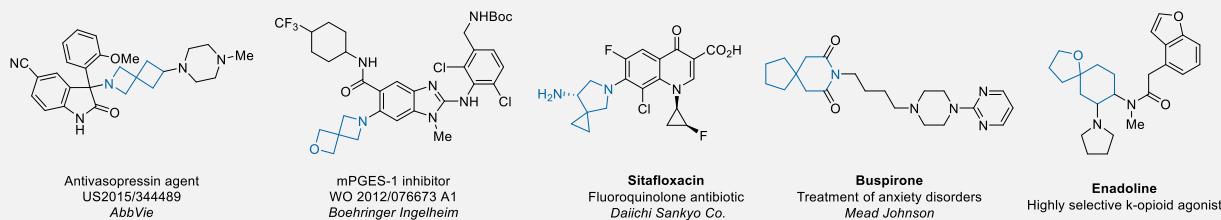
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Unique 3D-shaped Spirocycles to Explore Novel Chemical Space

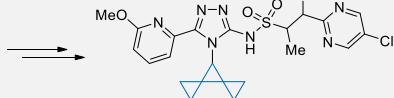
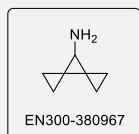
Introduction

Conformational rigidification of flexible compounds by introducing a ring is a popular strategy in drug design. The resulting cyclic analogues usually have a reduced conformational entropy penalty upon binding to a protein target. A conformational restriction can also be imposed by introduction of a spirocyclic ring. Spirocyclic systems are 3D-shaped, in strict contrast to flatten benzene compounds.¹⁻⁶ It is especially true for polycyclic compounds. In this context, *Enamine* offers a library of innovative three-cyclic scaffolds for drug design.

Unique Spirocycles in drug discovery



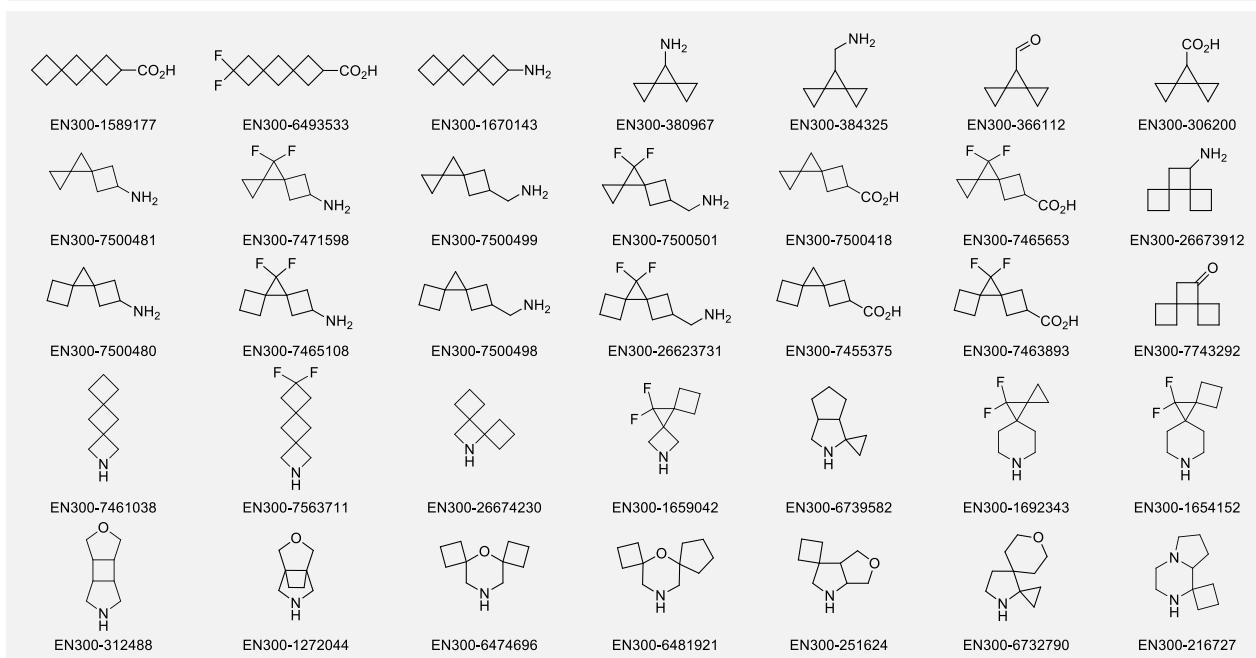
Case studies



Unique properties:

- 3D-shaped;
- Fsp³-rich;
- conformationally restricted;
- functionalization of pharmacophores;
- entry into novel chemical space.

We offer: more than 50 of three-cyclic building blocks from stock on a 5-10 g scale



References

1. Y. Zheng et al. *Bioorg. Med. Chem. Lett.* **2014**, *24*, 3673.
2. Y. Skalenko et al. *J. Org. Chem.* **2018**, *83*, 6275.
3. R. Bychek et al. *Chem. Eur. J.* **2018**, *24*, 12291.
4. A. Kirichok et al. *Angew. Chem. Int. Ed.* **2017**, *56*, 8865.
5. P. Nosik et al. *Adv. Synth. Catal.* **2017**, *359*, 3126.
6. Y. M. Sokolenko et al. *J. Org. Chem.* **2019**, *84*, 13908.



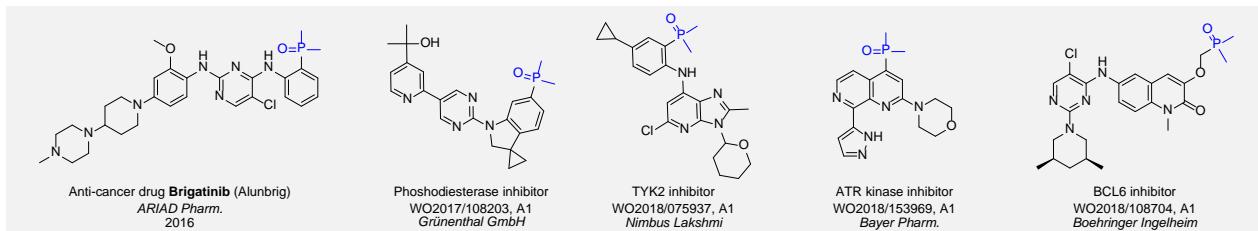
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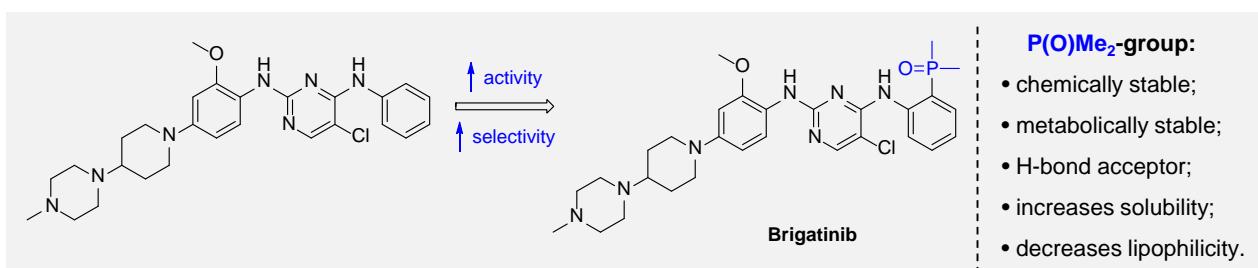
P(O)Me₂-containing Building Blocks for Drug Design

Introduction

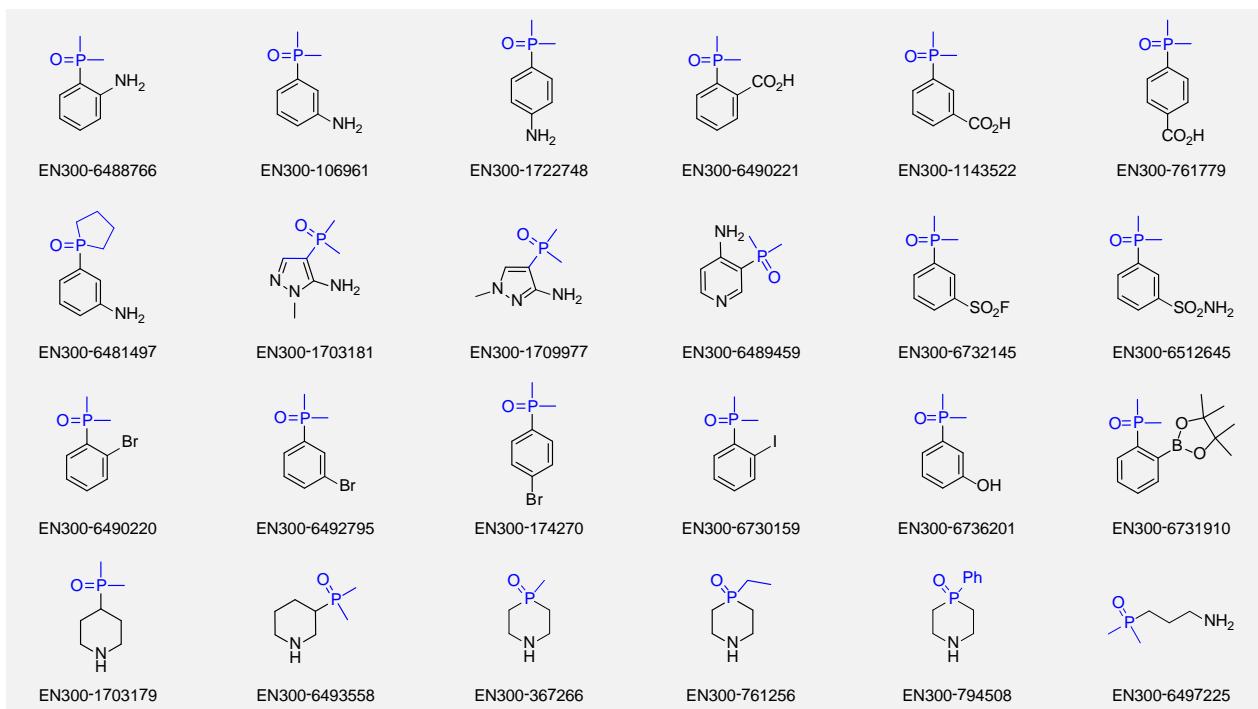
Phosphine oxides belong to a chemical class seldom employed in drug design. However, the FDA-approval of *Brigatinib* drug (ARIAD Pharm.) in 2017 may further inspire application of this unique functional group in medicinal chemistry. The highly ionic P=O bond imparts a number of important drug-like properties, including decreased lipophilicity, increased aqueous solubility, H-bond acceptor ability, and high metabolic stability.¹⁻³ Herein we have designed and synthesized a library of phosphine oxide derivatives for drug design.



Discovery of *Brigatinib*



We offer >30 unique P(O)Me₂-containing derivatives on a 5-50 g scale from our stock.



References

1. W.-S. Huang et al. *J. Med. Chem.* **2016**, *59*, 4948.
2. A. A. Kamel. *International Journal of Chemical and Biomedical Science*, **2015**, *1*, 56.
3. V. Iaroshenko. *Organophosphorus Chemistry: From Molecules to Applications*, John Wiley & Sons, **2019**, 568.

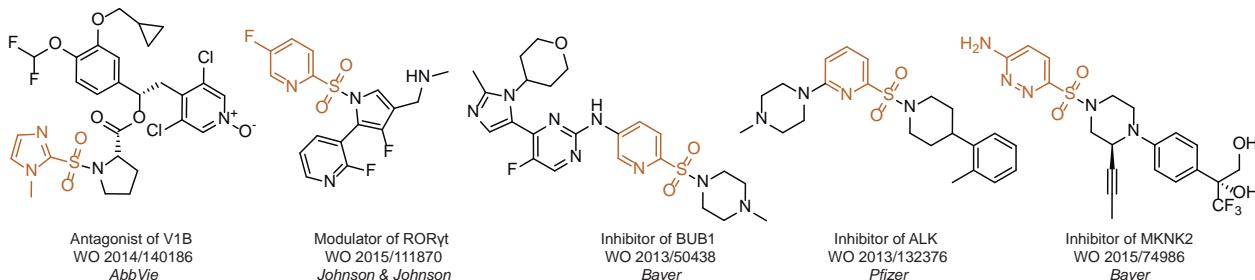


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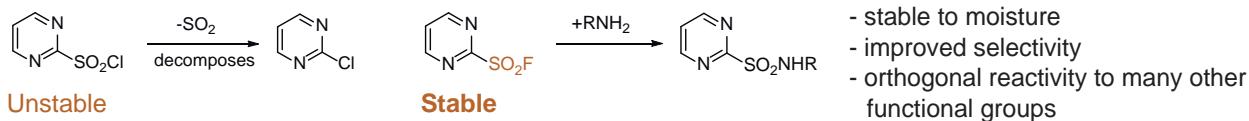
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SULFONYL FLUORIDES (-SO₂F): MORE OPTIONS FOR DRUG DESIGN

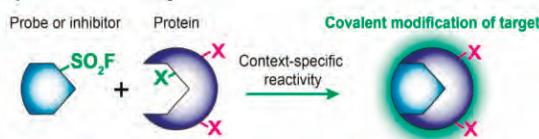
Sulfonyl chlorides (-SO₂Cl) are widely used in medicinal chemistry and agrochemistry as precursors to pharmacologically important sulfonamides. Many sulfonyl chlorides with heteroaromatic substituents, however, are unstable due to SO₂ extrusion. More stable sulfonyl fluorides (-SO₂F) in many cases are the only option to synthesize the desired sulfonamides. They are less reactive, so that they might even have a free aliphatic amino groups in their structure. Besides unique monofunctional sulfonyl fluorides, Enamine offers a wide array of scaffolds and linker compounds.



Properties of sulfonyl fluorides

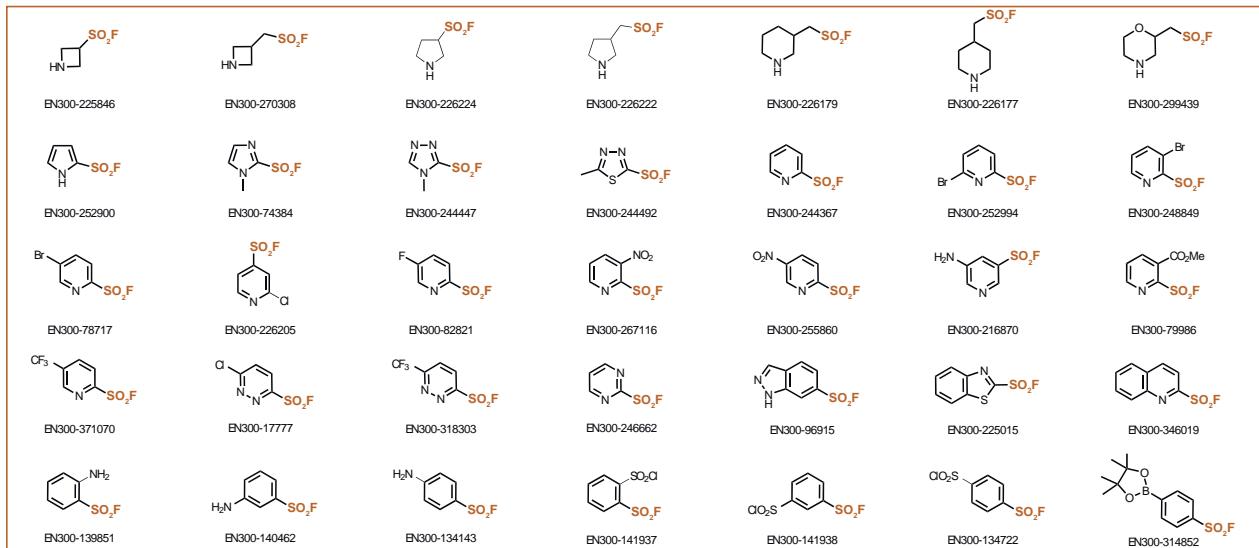


-SO₂F probes in chemical biology



The SO₂F group covalently binds to the residues of serine, threonine, tyrosine, lysine, cysteine, and histidine in proteins. Sulfonyl fluorides are widely used as chemical probes and covalent protein inhibitors.

Our offer: >200 Sulfonyl fluorides (-SO₂F) in gram amounts in stock
Custom synthesis of further analogues and compound libraries



References

- ¹ A. Narayanan *et al.* *Chem. Sci.* **2015**, 2650.
² J. Dong *et al.* *Angew. Chem. Int. Ed.* **2014**, 9430.
³ A. Garcia-Rubia *et al.* *Angew. Chem. Int. Ed.* **2011**, 10927.
⁴ S. W. Wright *et al.* *J. Org. Chem.* **2006**, 1080.
⁵ S. Caddick *et al.* *Org. Lett.* **2002**, 2549.

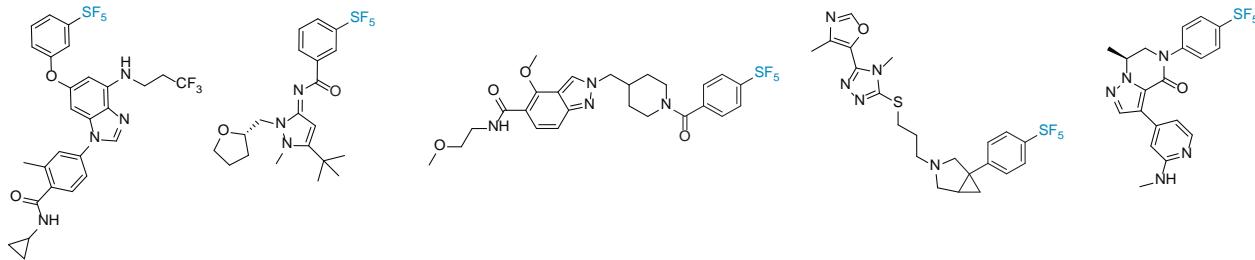


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SF₅-BUILDING BLOCKS

The organic chemistry of the pentafluorosulfanyl group (SF₅) has been developing since 1950's. As the SF₅ group is larger and more lipophilic than the CF₃ one, it is often considered as a "super-trifluoromethyl group". Over the past decade, the SF₅-containing aromatic compounds have found great practical application in medicinal chemistry.



pKa EtOH/H ₂ O 50:50	4.60	4.82	5.11	5.15	5.16	5.28
Lipophilicity (π) of substituent X						
X	SCF ₃	SF₅	OCF ₃	CF ₃	F	H
π_p	1.44	1.23	1.04	0.88	0.14	0

Our offer: >30 SF₅-buiding blocks in gram amounts in stock. Custom synthesis of further analogues and compound libraries

EN300-29892	EN300-129955	EN300-136698	EN300-131367	EN300-204011
EN300-67610	EN300-135382	EN300-131608	EN300-130393	EN300-154743
EN300-254687	EN300-251966	EN300-243790	EN300-130400	EN300-1704475

References

- ¹ R. Paul *et al. Chem. Rev.* **2015**, 1130.
² S. Altomonte *et al. J. Fluor. Chem.* **2012**, 57.

- ³ P. Kirsch. *Modern Fluoroorganic Chemistry*. **2004**, 146.



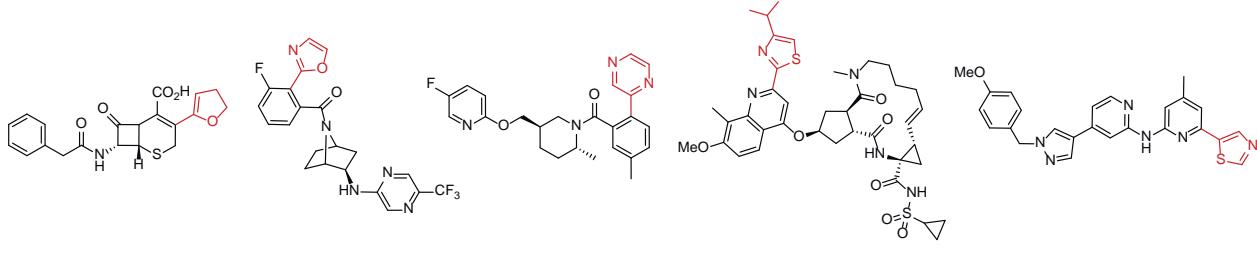
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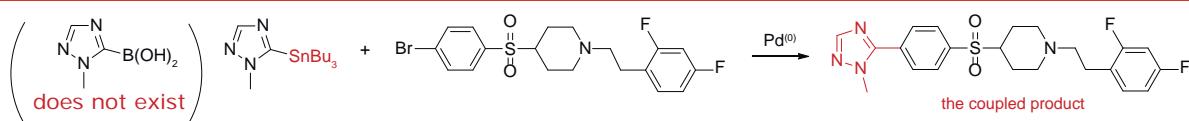
Stannanes for Drug Design

Introduction

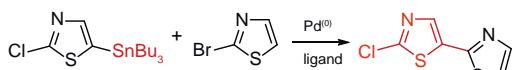
The Stille reaction has become one of the most powerful synthetic tools in organic chemistry. The Stille coupling as a versatile C-C bond forming reaction between stannanes and halides or pseudohalides, has very few limitations on the R-groups. Today, the Stille reaction constitutes a reliable and often-used method for the construction of carbocyclic and heterocyclic rings. Stannanes are stable and allow to prepare alternatives to unstable boronic acids or to be used in click chemistry.



Advantages



Reactivity

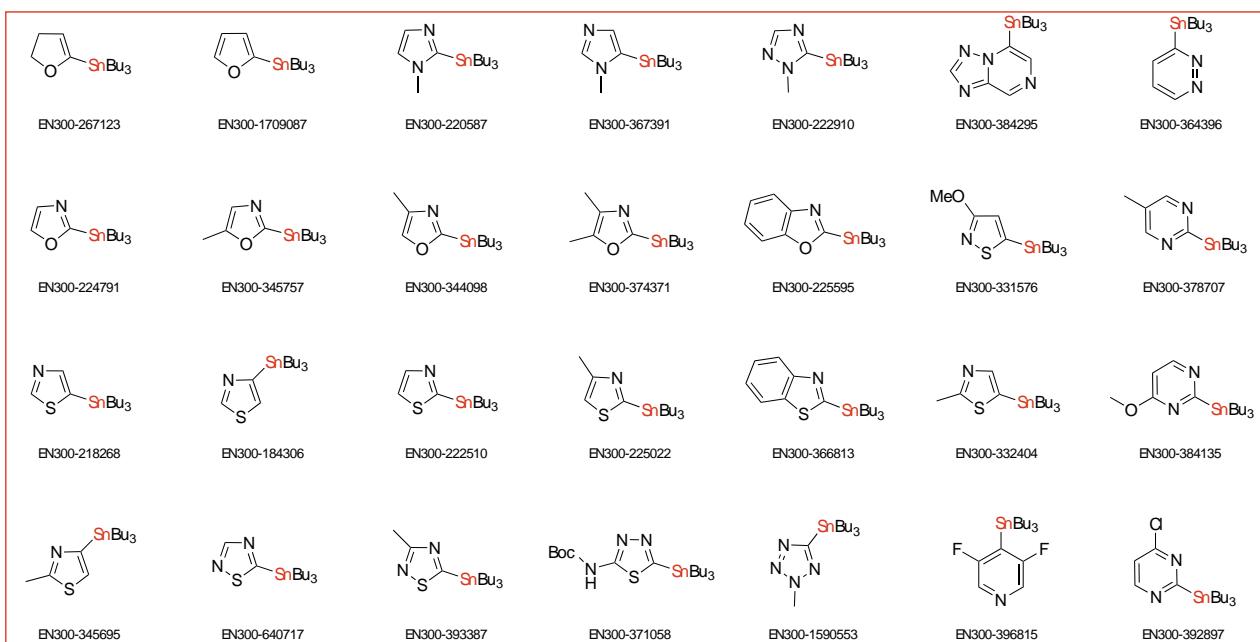


Synthesis of bithiazoles using Pd-catalyzed reactions in which Stille is superior to Negishi and Suzuki couplings.

Properties

- readily prepared, purified and stored;
- tolerate a wide variety of functional groups;
- require mild reaction conditions;
- not sensitive to moisture or oxygen;
- allow to prepare alternative compounds to unstable boronic acids

Our offer



References

- 1 C. Cordovilla, *ACS Catal.* **2015**, 3040.
- 2 M. M. Heravi *et al.* *Tetrahedron*. **2014**, 7.

- 3 K. C. Nicolaou *et al.* *Angew. Chem. Int. Ed.* **2005**, 4442.
- 4 O. Krebs *et al.* *Org. Lett.* **2005**, 1063.

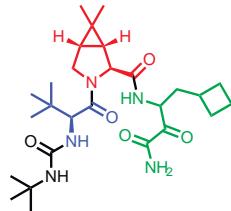


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antiviral drug

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- Custom synthesis of amino acids, their derivatives, and compound libraries

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Amino acids with polar and hydrophilic side chains						
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Amino acids with hydrophobic side chains						
EN300-114441	EN300-58167	EN300-34663	EN300-762994	EN300-344208	EN300-182109	EN300-178651
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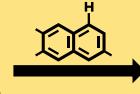
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